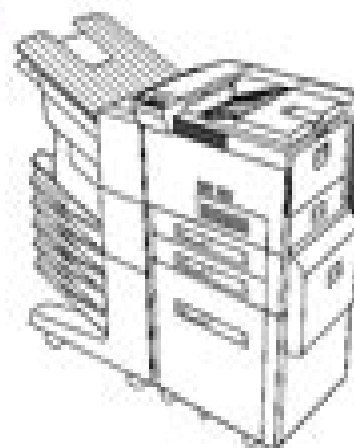




English



HP LaserJet 8100, 8100 N, 8100 DN
and Paper Handling Devices
Service Manual

Hp Laserjet 8100 8100n 8100dn Service Manual

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web start with 8.3 mole H_2S and multiply it with 2 mole H_2O over 2 mole H_2S from the equation $8.3\text{ mole H}_2\text{S} \times \frac{2\text{ mole H}_2\text{O}}{2\text{ mole H}_2\text{S}} \times 18\text{g H}_2\text{O} = 149.4\text{g H}_2\text{O}$ is your theoretical yield so percent yield $\frac{137.1\text{g H}_2\text{O}}{149.4\text{g H}_2\text{O}} \times 100 = 91.77\%$ problems 6 8 are similar to problem 5

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