

GENERAL CERTIFICATE OF EDUCATION BOARD

June 2022 GCE Examination Timetable

ORDINARY LEVEL		DATE	ADVANCED LEVEL	
MORNING SESSION (BEGINNING AT 8:00 A.M.)	AFTERNOON SESSION (BEGINNING AT 1:00 P.M.)		MORNING SESSION (BEGINNING AT 8:00 A.M.)	AFTERNOON SESSION (BEGINNING AT 1:00 P.M.)
0525. Economics 1 (25 N)	0525. Economics 2 (25 N)	TUESDAY 26/06/2022	0725. Economics 1 (25 N) 0755. Geology 1 (25 N)	0755. Geology 2 (25 N) 0725. Economics 2 (25 N)
0540. Biology 1 (25 N)	0540. Biology 2 (25 N)	WEDNESDAY 29/06/2022	0720. Biology 1 (25 N)	0750. History 1 (25 N) 0720. Biology 2 (25 N)
0550. Geography 1 (25 N)	0550. Geography 2 (25 N)	THURSDAY 30/06/2022	0750. Physics 1 (25 N) 0750. Geography 1 (25 N)	0725. Economics 2 (25 N) 0750. Physics 2 (25 N)
0555. Chemistry 1 (25 N) 0525. Literature in English 1 (25 N) 0540. History 1 (25 N) 0545. Human Biology 1 (25 N)	0555. Chemistry 2 (25 N) 0525. Literature in English 2 (25 N) 0540. History 2 (25 N) 0545. Human Biology 2 (25 N)	FRIDAY 01/07/2022	0750. History 2 (25 N) 0725. Chemistry 1 (25 N) 0740. Food Science & Nutrition 1 (25 N) 0750. Geography 2 (25 N)	0750. English Language 1 (25 N) 0725. Chemistry 2 (25 N) 0750. History 1 (25 N)
0550. English Language 1 (25 N) 0550. English Language 2 (25 N)		MONDAY 04/07/2022	0750. English Language 2 (25 N)	0740. French 1 (25 N) 0740. Special B.E. Education French 1 (25 N)
0542. Citizenship Education 1 (25 N)	0542. Citizenship Education 2 (25 N)	TUESDAY 05/07/2022	0750. Accounting 1 (25 N) 0725. Further Math 1 (25 N) 0750. Philosophy 1 (25 N)	0750. Geography 2 (25 N) 0740. Food Science & Nutrition 2 (25 N)
0545. French 1 (25 N) 0545. French 2 (25 N) 0545. Special Bilingual Education French 1 (25 N) 0545. Special Bilingual Education French 2 (25 N)	0555. Computer Science 1 (25 N) 0555. Computer Science 2 (25 N)	WEDNESDAY 06/07/2022	0745. French 1 (25 N) 0745. Special B.E. Education French 2 (25 N)	0745. French 2 (25 N) 0745. Special B.E. Education French 2 (25 N)
0550. Commerce 1 (25 N) 0540. Physics 1 (25 N)	0550. Commerce 2 (25 N) 0540. Physics 2 (25 N)	THURSDAY 07/07/2022	0745. Math with Mechanics 1 (25 N) 0750. Math with Statistics 1 (25 N) 0725. Literature in English 2 (25 N) 0745. Math with Mechanics 2 (25 N) 0750. Math with Statistics 2 (25 N) 0725. Literature in English 2 (25 N)	0745. Math with Mechanics 2 (25 N) 0750. Math with Statistics 2 (25 N) 0725. Literature in English 2 (25 N) 0750. Accounting 2 (25 N) 0725. Further Math 2 (25 N) 0750. Philosophy 2 (25 N)
0570. Mathematics 1 (25 N)	0570. Mathematics 2 (25 N)	FRIDAY 08/07/2022	0750. Computer Science 1 (25 N) 0750. Info and Comm. Tech 1 (25 N) 0750. Religious Studies 1 (25 N) (from 08:00 am)	0750. Accounting 2 (25 N) 0725. Further Math 2 (25 N) 0750. Philosophy 1 (25 N)
0555. Accounting 1 (25 N) 0550. Logic 1 (25 N) 0555. Geology 1 (25 N) 0575. Additional Math 1 (25 N) 0540. Food & Nutrition 1 (25 N)	0555. Accounting 2 (25 N) 0550. Logic 2 (25 N) 0555. Geology 2 (25 N) 0575. Additional Math 2 (25 N) 0540. Food & Nutrition 2 (25 N)	SATURDAY 09/07/2022	0750. Religious Studies 2 (25 N)	0750. Computer Science 2 (25 N) 0725. Further Math 2 (25 N) 0750. Philosophy 2 (25 N)
0545. Religious Studies 1 (25 N) 0545. Religious Studies 2 (25 N)		TUESDAY 12/07/2022	0750. Religious Studies 1 (25 N)	

PRACTICAL TIMETABLE: - 0740 SEE French: Tuesday 07/07/2022, 0540 Food and Nutrition and 0725 Chemistry: Tuesday 07/07/2022 - Friday 09/07/2022, 0755 Computer Science: Friday 10/07/2022, 0750 Physics: Monday 13/07/2022 - Wednesday 15/07/2022, 0545 SEE French: Tuesday 14/07/2022, 0755 Geology: Tuesday 14/07/2022, 0750 ICT: Thursday 16/07/2022 - Friday 17/07/2022, 0718 Biology: Thursday 16/07/2022 - Tuesday 21/07/2022, 0555 Computer Science: Tuesday 21/07/2022 - Thursday 23/07/2022, 0740 Food Science and Nutrition: Wednesday 23/07/2022 - Friday 24/07/2022.

- NOTE: Where the dates / days of the examination conflict with any other examination, event or document, this timetable shall prevail, except for changes specifically announced by the Board.
- Chiefs of schools and candidates should carefully study the Timetable before making their entries in order to avoid clashes.



Gce Ecz Exam Time Table 2015

TD Snyder



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In a world taken by displays and the ceaseless chatter of fast connection, the melodic beauty and mental symphony developed by the written term frequently disappear in to the background, eclipsed by the relentless noise and interruptions that permeate our lives. But, situated within the pages of **Gce Ecz Exam Time Table 2015** a wonderful literary prize brimming with natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, this captivating masterpiece conducts readers on a mental journey, well unraveling the hidden songs and profound impact resonating within each cautiously crafted phrase. Within the depths of this moving examination, we will examine the book is main harmonies, analyze their enthralling publishing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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web hydrogen will be left slightly positive so this is a polar molecule i represent this as a couple of different animals here got a bear and it s got the five valence electrons and it shares the single electrons with its penguin friend the penguins each have one valence electron to be shared with from the bear

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web what does the crossed arrow represent in the comic strip 11 what are two of the definitions of dipole given in the comic strip 12 making sense 13 if you finish early what does electronegativity have to do with polarity using polar bears and penguins create an illustration showing a hydrogen sulfide molecule H_2S

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of H_2O on page 4 they are being pulled towards the stronger polar bear he s more electronegative 11

explain what you think the crossed arrow represents in the comic book polar bonds perhaps 12

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web lesson 3 3 3 polar bears and penguins 43 lesson 3 3 4 thinking electronegatively 50 this unit could also be called organic chemistry which includes studying the kinds of compounds that are essential for living organisms answer the following question examine the following molecules what patterns do you see in the bonding of

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web the two polar bears or penguins are equal in electronegativity therefore they pull electrons equally as hard and neither one will end up with more electrons more of the time 4 a what 3 types of bonds are represented on the third page of the comic book nonpolar covalent polar covalent and ionic b

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web answer because the two polar bears or penguins have equal electronegativity they will pull electrons equally hard and in most cases neither will end up with more electrons each polar bear and penguin has an electronegativity of the same value a non polar bond is formed when two polar bears or two penguins are bonded together

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web why does the polar bear tell the penguin to try to be more positive the polar bear benefits more when the penguin is more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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web answer the following questions consider the following illustration draw the lewis dot structure for hcl if the penguin represents a hydrogen atom and the polar bear represents a chlorine atom what does the ice cream represent in the drawing what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web explain how the iceberg penguins and polar bears represent trends in electronegativity on the periodic table and more study with quizlet and memorize flashcards containing terms like how does the comic define a polar molecule define

electronegativity as you understand it after reading the first two pages of the comic interpret the pic
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web so the final representation of the NH_3 molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the NH_3 molecule

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web question using polar bears and penguins create an illustration showing an ammonia molecule NH_3 hint you may wish to start with a lewis dot structure

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges

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