

Show all work; observe significant figures and units. Remember to predict first!

1. A balloon with a volume of 5.0 L at 1.0 atm is placed in a vacuum pump. When the pump is turned on the pressure drops to 0.25 atm.
 - a. Will the volume of the balloon get bigger or smaller?
 - b. Calculate the new volume of the balloon.

2. A man who is standing in a valley where the pressure is 1.1 atm blows a balloon until it has a volume of 5 L. He then takes the balloon up to a high mountain peak where the pressure is 0.85 atm.
 - a. Will the volume of the balloon get bigger or smaller?
 - b. Calculate the new volume of the balloon.

3. A beach ball with a volume of 10.2 L at 19°C is taken outside where the temperature is 30°C.
 - a. Will the volume of the ball get bigger or smaller?
 - b. Calculate the new volume of the ball.

4. A balloon with a volume of 3.3 L at room temperature shrinks to 3.0 L when it is placed in an ice bath (temp = 0°C).
 - a. Is the room temperature higher or lower than the ice bath?
 - b. Calculate the room temperature.

5. The pressure of a gas at 75°C is 852 mm Hg. When the temperature is lowered to 12°C,
 - a. Will the pressure go up or down?
 - b. Calculate the new pressure.

6. A pot roast is cooking in a pressure cooker at 177°C. Room conditions are 20°C and 1 atm.
 - a. Is the pressure inside the pressure cooker higher or lower than room pressure?
 - b. Calculate the pressure inside the pressure cooker.

Gas Laws Worksheets With Answers

**Robert C. Smoot, Smoot, Richard G.
Smith, Jack Price**



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