

Gas Laws Problem Packet

Gas Laws Packet

Name _____ Date _____ Period _____

Introduction

Standard Conditions

Unit	Standard Pressure
atm	1.013
mmHg	760
cmHg	76
torr	760

22.4 L = 1 mol at STP

• STP (standard temperature and pressure) is 0°C and 1 atm.

• The gas laws:

Gas Law	Equation	Variables
Boyle's Law	$P_1V_1 = P_2V_2$	P, V
Charles's Law	$\frac{V_1}{T_1} = \frac{V_2}{T_2}$	V, T
Gay-Lussac's Law	$\frac{P_1}{T_1} = \frac{P_2}{T_2}$	P, T
Avogadro's Law	$\frac{V_1}{n_1} = \frac{V_2}{n_2}$	V, n
Combined Gas Law	$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$	P, V, T
Ideal Gas Law	$PV = nRT$	P, V, n, T

• The constant R used in the ideal gas law depends on the unit of pressure.

a. Units: $R = 0.0821 \frac{\text{atm} \cdot \text{L}}{\text{mol} \cdot \text{K}}$

b. Units: $R = 62.36 \frac{\text{mmHg} \cdot \text{L}}{\text{mol} \cdot \text{K}}$

Notes:

- Temperatures must always be in Kelvin: $K = ^\circ\text{C} + 273$.
- If you are given two different units of pressure, convert first.
- Circle two columns before solving. Circle what you're given in the problem. Keep the two that use the variables you're given, and solve for the unknown. For ideal gas laws, write what you are given, insert those in equation and solve for unknown.
- Avogadro's Law: Solve for volume to use ideal standard values.
- Charles's Law: You will usually have the molecular mass for these problems. Double and triple check you have the right of effluent and molecular masses in the correct units before solving the equation (this is an important relationship). You must evaluate the answer and compare both sides of the equation before moving things.

Charles' Law $V_1/T_1 = V_2/T_2$

1. The temperature of a gas is directly proportional to its volume. If the temperature of a gas is doubled, the volume will also double. If the temperature of a gas is halved, the volume will also halve.

2. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

3. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 2.0 atm?

4. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 0.5 atm?

5. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.5 atm?

6. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

7. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

8. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

9. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

10. A gas at 100°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 200°C and 1.0 atm?

All Gas Laws Included!

Gas Laws Packet Answers

Gerald R. North, Tatiana L. Erukhimova

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Proposed Changes to Natural Gas Laws United States. Congress. House. Committee on Energy and Commerce. Subcommittee on Fossil and Synthetic Fuels, 1983

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Atmospheric Thermodynamics Gerald R. North, Tatiana L. Erukhimova, 2009-04-09 This textbook presents a uniquely integrated approach in linking both physics and chemistry to the study of atmospheric thermodynamics The book explains the classical laws of thermodynamics focuses

on various fluid systems and recognising the increasing importance of chemistry in the meteorological and climate sciences devotes a chapter to chemical thermodynamics which includes an overview of photochemistry Although students are expected to have some background knowledge of calculus general chemistry and classical physics the book provides set aside refresher boxes as useful reminders It contains over 100 diagrams and graphs to supplement the discussions and a similar number of worked examples and exercises with solutions included at the end of the book It is ideal for a single semester advanced course on atmospheric thermodynamics and will prepare students for higher level synoptic and dynamics courses

Instructional Technology Research, Design and Development: Lessons from the Field Alias, Nor

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Learning and Leading with Technology, 1996

Applied Thermodynamics for Meteorologists Sam Miller, 2015-06-04 This textbook on atmospheric thermodynamics is for students of meteorology or atmospheric science It also serves as a reference text for working professionals in meteorology and weather forecasting It is unique because it provides complete calculus based derivations of basic physics from first principles and connects mathematical relationships to real world practical weather forecasting applications Worked examples and practice problems are included throughout

Oil and Gas Law: 1922-1951
Texas Law Review, 1954

Law Day U.S.A. Planning Guide and Program Manual American Bar Association, 1968

Water: Fundamentals as the Basis for Understanding the Environment and Promoting Technology P.G. Debenedetti, M.A. Ricci, F. Bruni, 2015-05-18 Water fascinating in all its phases forms and states of aggregation Without it life as we know it would not exist on Earth for as Paracelsus stated in the 16th century water is the matrix of the world and of all its creatures While it may appear to be a simple molecule there is still much about it which is not fully understood What is notably lacking is a microscopically based understanding of the reasons for the many anomalous properties of water This book presents lectures from the Enrico Fermi summer school Water fundamentals as the basis for understanding the environment and promoting technology held in Varenna Italy in July 2013 The aim of the school was to offer a glimpse of the many questions that remain unanswered about this molecule and topics covered included water in relation to other liquids biological water local environment of water protons atmospheric water amorphous solid phases of water NMR studies of water spectroscopic studies of water the structure of liquid water and supercooled water among others While this list is by no means exhaustive or complete it is wide enough to provide a solid basis to young researchers in the field and the book will be a valuable source

of reference for students and all those with an interest in the properties of this fascinating substance **United States Government Organization Manual** ,1969 **Terrestrial Hydrometeorology** W. James Shuttleworth,2012-01-30 Both hydrologists and meteorologists need to speak a common scientific language and this has given rise to the new scientific discipline of hydrometeorology which deals with the transfer of water and energy across the land atmosphere interface Terrestrial Hydrometeorology is the first graduate level text with sufficient breadth and depth to be used in hydrology departments to teach relevant aspects of meteorology and in meteorological departments to teach relevant aspects of hydrology and to serve as an introductory text to teach the emerging discipline of hydrometeorology The book will be essential reading for graduate students studying surface water hydrology meteorology and hydrometeorology It can also be used in advanced undergraduate courses and will be welcomed by academic and professional hydrologists and meteorologists worldwide Additional resources for this book can be found at <http://www.wiley.com/go/shuttleworth>

hydrometeorology **Resources in Education** ,1993 **Texas Oil and Gas Law Journal** ,2002 *Meteorology* ,1971

A First Course in Atmospheric Thermodynamics Grant W. Petty,2008-01-01 This textbook is written for meteorology majors who require an initial introduction to the physical properties of the atmosphere and to the essential principles and real world applications of atmospheric thermodynamics These topics are supplemented by a sampling of techniques and technologies related to atmospheric measurements and observations A unique tutorial included as an appendix teaches students how to attack physical problems symbolically deferring numerical calculations until the final step in the solution The author's objectives include not only covering the traditional core subject matter of an undergraduate thermodynamics course but also facilitating students transition from a purely abstract understanding of calculus and physics concepts to the confident application of both to the science of meteorology **Examining Impacts of Federal Natural Resources Laws Gone Astray** United States. Congress. House. Committee on Natural Resources. Subcommittee on Oversight and Investigations,2017 **Examining Impacts of Federal Natural Resources Laws Gone Astray, Part II, Serial No. 115-16, July 18, 2017, 115-1** ,2017 **Business Law** Dr. S. K. Singh, Dr. Angad Tiwary, Nikhil Gupta,2021-05-31 Unit I Indian Contract Act 1872 1 Business Mercantile Law An Introduction 2 Indian Contract Act 1872 An Introduction 3 Contract Meaning Definition and Characteristics of a Valid Contract 4 Agreement Meaning Kinds and Difference 5 Proposal Offer Acceptance Communication and Revocation 6 Capacity of Parties to Contract or Parties Competency to Contract 7 Free Consent 8 Lawful Consideration and Objects 9 Agreements Expressly Declared as Void 10 Contingent Contracts 11 Performance of Contracts and Appropriation of Payments 12 Discharge of Contracts 13 Quasi or Implied Contracts of Certain Relations Resembling those Created by Contracts Sections 68 to 72 14 Remedies for Breach of Contract Unit II Special Contracts 15 Contract of Indemnity and Guarantee 16 Contract of Bailment and Pledge 17 Contracts of Agency Unit III The Sale of Goods Act 1930 18 The Sale of Goods Act 1930 An Introduction 19 Conditions and Warranties 20 Effects of the

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