

## Gas Laws & SCUBA Diving

Read the accompanying article "Gas Laws and SCUBA Diving," ChemMatters, February 1993, pp.24.

Answer the following questions completely and upload in the assignment area.

- a. What is gas? name and describe?
- b. Why does diving 30 meters below sea level affect our bodies more than being in a building 30 meters above sea level?
- c. What parts of a diver's body are most affected by pressure changes?
- d. What Boyle's Law
- e. Why does a SCUBA diver's lungs collapse as they ascend?
- f. What would happen to a diver who ascends too quickly without having been a 30-meter diver? Explain in terms of Boyle's Law
- g. What Henry's Law
- h. What gas is associated with a warning bubble for this reason and what does it do?
- i. What is another reason for decompression sickness?
- j. Explain how these decompression factors affect a diver's symptoms upon ascent and descent.
- k. What is nitrogen narcosis?
- l. What is gas law nitrogen rule and what does it (What does ascending off 100' with much nitrogen cause that fatal air embolism to happen? Explain.
- m. What relationship between the composition of gases and the solubility of a gas is it all gas or inert gas relationships? Explain.
- n. What gas causes the BEE for people who in the atmosphere but is different for those at sea level?

# Gas Laws And Scuba Diving Chemmatters Answers

**M Mosston**



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