

and scattering is *incorrect*:

vely charged object .

increase in the impact parameter .

Coulomb interaction with the outer valence 4-d

es are deflected through 45° or more by a $0.1 \mu\text{m}$ gold

A MILLIA ISLAMIA

2019-2020

obtained in the H_α emission spectrum from hydrogen

λ) of an atomic spectral line of wavelength 500 nm

correctly describes a system of two identical, non-
and 'b' ?

$$\psi_A = 1/2[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$$

$$= [\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$$

$$\psi_A = 1/\sqrt{2}[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$$

$$\psi_A = 1/\sqrt{2}[\psi_a(1)\psi_b(2) + \psi_a(2)\psi_b(1)]$$

described by the term symbol $3^2S_{1/2}$. What are the
e valence electron?

65. Which of the following statements

(A) Molecular rotational states are

(B) The selection rule for molecu-

(C) Hydrogen molecule (H_2) does

(D) Transitions between low lying

887 1A

66. A 50 kW broadcasting antenna em
photons are emitted per second are

(A) 1.5×10^{31}

(B) 1.7×10^{31}

(C) 1.9×10^{31}

(D) 2×10^{31}

67. A photon is incident on an electros
observed to be 5 times the wavelength
the wavelength of the incident photon

(A) $2 \times 10^{-13} \text{m}$

(B) $3 \times 10^{-13} \text{m}$

(C) $4 \times 10^{-13} \text{m}$

(D) $5 \times 10^{-13} \text{m}$

68. The de Broglie wavelength of a pa
box of length L, is

(A) $4L$

(B) $2L$

(C) L

(D) $L/2$

69. The operator $\left(\frac{d}{dx} + x\right)\left(\frac{d}{dx} - x\right)$ is

(A) $\frac{d^2}{dx^2} - x^2$

(B) $\frac{d^2}{dx^2} - x^2 - 1$

(C) $\frac{d^2}{dx^2} - 2x \frac{d}{dx} + 1$

(D) $\frac{d^2}{dx^2} - 2x \frac{d}{dx} - x^2$

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