



Delhi Public School, Prayagraj

HALF YEARLY EXAMINATION (2022-2023)

SUBJECT: - CHEMISTRY

CLASS: - XI

Duration: 3 Hours

Maximum Marks: 70

Name: - ~~XXXXXXXXXX~~

Class/Sec.: - ~~XI/A~~

Instructions:

Q.No. 1 to 10 carries 1 mark each.

Q.No. 11 to 19 carries 2 marks each

Q.No. 20 to 28 carries 3 marks each

Q.No. 29 to 31 carries 5 marks each.

- Which of the following compounds has hydrogen with oxidation number +1?
(a) HBr (b) NaOH (c) LiH (d) H₂
- In the compounds KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$, the highest oxidation state is of the element:
(a) Mn (b) K (c) Cr (d) O
- If travelling at same speeds, which of the following matter waves have shortest wavelength?
(a) Proton (b) Neutron (c) Alpha particle (d) Electron
- Which of the following pairs of gases contains the same number of molecules?
(a) 16 gm of O_2 and 14 gm of N_2 (b) 8 gm of O_2 and 22 gm of CO_2
(c) 28 gm of N_2 and 22 gm of CO_2 (d) 32 gm of O_2 and 32 gm of N_2
- The number of water molecules present in a drop of water (volume 0.0018 ml) at room temperature is:
(a) 6.023×10^{23} (b) 1.084×10^{24} (c) 4.84×10^{23} (d) 6.023×10^{23}
- Which of the following is true?
(a) According to Pauli's exclusion principle, no two electrons in an atom can have the same values of quantum numbers n, l and m.
(b) The total energy of an electron in an orbital is half of its potential energy.
(c) The speed of an electron in an orbit increases with increase of quantum number n.

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