

# Answer key

## Unit 1

### 1a (pages 4 and 5)

- 1  
1 works 2 specialises 3 goes 4 studies 5 wants  
6 spends 7 don't stay 8 travel 9 visit 10 don't realise  
11 helps 12 don't have
- 2  
1 starts 2 watches 3 flies 4 passes 5 lives  
6 studies 7 finishes 8 relaxes
- 3  
/s/ helps, visits, wants  
/s/ has, is, spends, stays, studies, travels  
/rɪz/ realises, specialises
- 4  
1 Where does Nathan work?  
2 Where does he often go?  
3 What does he find and study?  
4 Where does he spend a lot of time?  
5 Why do new viruses travel more easily?  
6 What does he need for his work?  
7 Do many people have electricity and running water?  
8 How does Nathan communicate?
- 5  
1 c 2 b 3 a 4 a
- 6  
1 I always do exercise in the evening.  
2 It is always colder in the winter.  
3 I take this medicine twice a day.  
4 They don't often go on holiday.  
5 We are sometimes busy at weekends.  
6 She rarely eats out during the week.  
7 You are never on time for work.  
8 Do you always check your emails at lunchtime?

### 1b (pages 6 and 7)

- 1  
1 b 2 c 3 b 4 b 5 c 6 c 7 b 8 a
- 2  
1 patient 2 mothers-to-be 3 hospital 4 nutrition  
5 vaccinations 6 shortage 7 preventative 8 advice  
9 intensive training 10 ongoing
- 3  
Today they are visiting their first patient.  
Sarubai is checking Romi ...  
While they are checking the baby ...  
... and the number is growing.
- 4  
1 I'm driving  
2 has  
3 do you come  
4 is flying  
5 I never cycle  
6 is standing  
7 Do you understand  
8 it's getting  
9 aren't staying  
10 Are you working; are you taking
- 5  
a 5 b 4 c 3 d 3 e 4 f 4

- 6  
1 living 2 dropping 3 letting 4 swimming 5 having  
6 lying 7 taking 8 travelling 9 getting 10 jogging
- 7  
I usually get up at about seven o'clock and go running for half an hour. Then I feel ready for the day. I leave the house at about eight thirty and arrive at the hospital by nine. Currently, I'm seeing lots of children with flu. After work I often walk home. Sometimes friends come round for dinner, but I need eight hours of sleep a night so I'm always in bed by eleven o'clock.

### 1c (page 8)

- 1  
1 happy 2 money 3 students 4 concert
- 2  
1 c 2 b 3 b 4 c 5 a 6 a
- 3  
1 f 2 c 3 b 4 a 5 d 6 e
- 4  
1 e 2 d 3 a 4 c 5 b
- 5  
Possible answers:  
How do you feel?  
Do you feel like a coffee?  
Do you feel like doing something?  
How do you feel about pizza for dinner?  
Do you feel OK?  
What do you feel like doing?  
How does the weather feel?  
What's the weather like?

### 1d (page 9)

- 1  
1 nose 2 ear 3 tooth 4 head 5 throat 6 back  
7 mouth 8 stomach
- 2  
1 bad 2 ear 3 off 4 eat 5 see 6 play
- 3  
Tick the following for 1–3:  
1 sore throat, cough 2 high 3 pills  
4 Advice: Go to bed for two days. Drink lots of water.  
Come back if you still feel ill.
- 4  
1 How do you feel  
2 Let me have a  
3 Do you feel  
4 Have you got  
5 Let me check  
6 take this prescription  
7 You need to  
8 They are good  
9 try drinking  
10 If you still feel ill
- 5  
Model answers:  
You need to take some pills.  
Try taking some hot water with lemon and honey.  
Go to bed for a couple of days.  
You need to see a doctor.  
Take this medicine. It's good for flu.

## Kamico Assessment 8 English 2 Answer Key

**Patricia Healey, Patricia Healey, Ihm,  
Ma, Irene Kervick, Irene Kervick, Ihm,  
Ma, Anne B McGuire, Ihm, Ma, Anne B.  
McGuire, Adrienne Saybolt, Adrienne  
Saybolt, Ihm, Ma**

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web a model of acids and bases in which an acid is a hydrogen ion donor and a base is a hydrogen ion acceptor conjugate

acid the species produced when a base accepts a hydrogen ion from an acid

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web arrhenius model of acids a substance that contains hydrogen and ionizes to produce hydrogen ion in aqueous solution

arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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web kw the product of the concentrations of the hydrogen ions and hydroxide ions acidic solution one which  $\text{H}^+$  is greater than the  $\text{OH}^-$  basic solution one in which the  $\text{H}^+$  is less than the  $\text{OH}^-$  alkaline solutions the  $\text{H}^+$  of a basic solution is less than  $1 \times 10^{-7} \text{ M}$

they are the same as basic solutions

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web jan 10 2021 to identify the strongest base we can determine their weakest conjugate acid the conjugate acids of  $\text{CH}_3\text{NH}_2$  and  $\text{S}^{2-}$  are  $\text{CH}_4$   $\text{NH}_3$  and  $\text{HS}^-$  respectively next we consider that acidity increases with positive charge on the molecule thus ruling out that  $\text{S}^{2-}$  is the weakest base

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web compounds formed from combinations of acids and bases taste salty and are referred to in chemistry as salts sweet

compounds have characteristics of both acids and bases in the same molecule we will explore the relationship between

molecular structure and acids bases and consider water solutions of acids and bases

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web conjugate acid base pair two substances related by the loss or gain of a single hydrogen ion hydronium ion  $\text{H}_3\text{O}^+$  a water molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of electrons during a reaction while a base donates a pair of electrons

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web solutions manual acids i and bases section 18 1 introduction to acids and bases pages 634 643 practice problems pages 635 640 write balanced equations for reactions between the following aluminum and sulfuric acid  $2\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2$  calcium carbonate and hydrobromic acid