

## II PUC CHEMISTRY SYLLABUS BLOW-UP

### UNIT-I

#### Solid State

8 hrs

General characteristics of solids: amorphous and crystalline solids – examples, differences. Classification of crystalline solids – based on binding forces: molecular solids – (non-polar, polar, H-bonded), ionic solids, metallic solids, covalent or network solids – examples for all.

Definitions – lattice point, crystal lattice, unit cell, coordination number. Parameters of a unit cell, names of seven crystal systems, calculation of number of atoms in a cubic unit cell – simple cubic, fcc, bcc. Close packing in two-dimensional and three dimensional lattices – brief information, voids – types of voids, tetrahedral and octahedral and their relative numbers, calculation of the formula of the compounds based on the number of voids filled. Packing in solids – calculation of packing efficiency- fcc/bcc, bcc, simple cubic. Formula to calculate the density of the unit cell to be assumed- use the formula to calculate  $a$ ,  $d$ ,  $z$ ,  $M$ ,  $N_A$ . Numerical problems.

Point defects-types, a brief account of Frenkel and Schottky defects, metal excess defect and metal deficiency defect with examples.

Electrical properties: classification into conductors, insulators and semiconductors – their comparison based on band theory of metals, n-type and p-type semiconductors – differences and examples. Magnetic properties of substances – paramagnetism, diamagnetism and ferromagnetism, examples.

### UNIT-II

#### Solutions

9 hrs

Types of Solutions – binary – gaseous, liquid and solids, expressing the concentration of a solution of a solid in a liquid – mole fraction, molarity and molality. Solubility, solubility of a gas in a liquid – Henry's law, graph of partial pressure of a gas vs its mole fraction in solution, effect of pressure, temperature, applications of Henry's law.

Solution of liquid in liquid – Raoult's law- statement, mathematical expression, numerical problems, ideal solution – characteristics, graph, non-ideal solution –types – their characteristics and differences, examples, azeotropes – types, examples.

Solution of a solid in a liquid – Raoult's law – colligative properties – relative lowering of vapour pressure, elevation in boiling point, depression in freezing point, graphs for elevation in boiling point and depression in freezing point, SI units for  $K_b$ ,  $K_f$ , osmosis – osmotic pressure, isotonic, hypertonic, hypotonic solutions, reverse osmosis – application in desalination of water. Numerical problems on determination of molar mass using colligative properties.

Abnormal molar mass, van't Hoff factor  $i$ , value of  $i$  for non-electrolytes and solutes that associate or dissociate in dilute solution.

# Guidelines For Setting Ll Puc C

**Public Utilities Commission of Ohio**



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*Regulation of Radioactive Waste Management, Essays B1; Classification Systems for Radioactive Waste Disposal* ,1982

**Implementation of the Public Utility Regulatory Policies Act of 1978** United States. Congress. House. Committee on Interstate and Foreign Commerce. Subcommittee on Energy and Power,1980      **Decisions of the Public Utilities**

**Commission of the State of California** California Public Utilities Commission,1999-08-05      **Oregon Administrative**

**Rules** ,2001      **Implementation of the Public Utility Regulatory Policies Act of 1978** United States. Congress. Senate. Committee on Energy and Natural Resources,1986      Rules and Regulations U.S. Nuclear Regulatory Commission,1987

Code of Federal Regulations ,1970 Special edition of the Federal Register containing a codification of documents of general applicability and future effect with ancillaries      **Federal Energy Guidelines** United States. Federal Energy

Regulatory Commission,1990      *FCC Record* United States. Federal Communications Commission,2016      *Revised Final*

*Statement of Reasons, Modification of Regulations Establishing Enforcement Procedures for the Renewables Portfolio*

*Standard for Local Publicly Owned Electric Utilities* ,2016      **Federal Register** ,2014      **Federal Power Commission**

**Reports** United States. Federal Power Commission,1961 Contains all the formal opinions and accompanying orders of the Federal Power Commission In addition to the formal opinions there have been included intermediate decisions which have

become final and selected orders of the Commission issued during such period      **Rules, Regulations and Specifications**

Public Utilities Commission of Ohio,1929      **Nuclear Waste Disposal** United States. Congress. Senate. Committee on Environment and Public Works. Subcommittee on Nuclear Regulation,1980

## Adopting the Beat of Term: An Emotional Symphony within **Guidelines For Setting Ll Puc C**

In a global eaten by displays and the ceaseless chatter of instant conversation, the melodic splendor and mental symphony produced by the written word often disappear into the backdrop, eclipsed by the constant sound and distractions that permeate our lives. However, located within the pages of **Guidelines For Setting Ll Puc C** a charming fictional value brimming with fresh feelings, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that charming masterpiece conducts viewers on a psychological trip, well unraveling the hidden melodies and profound affect resonating within each cautiously crafted phrase. Within the depths of this touching examination, we will examine the book is main harmonies, analyze their enthralling writing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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