Thermochemistry and Gas Laws Unit Review

Energy Math

 How much energy is released when a brick with a mass of 329g is cooled from 80°C to 50°C? The specific heat capacity of brick is 0.84 1/g°C.

What is the mass of a hotdog that releases 3600 J of energy as it is cooled from 95°C to 45 °C?
 The specific heat capacity of a hotdog is 1.5 J/g°C.

 What is the specific heat capacity of a 350 g object that rises in temperature 15°C when 55,900 J of energy are added?

 45,920 J of energy are added to a 365g block of copper originally at 35°C. What will the new temperature of the copper be? The specific heat capacity of copper is 0.385 ½g°C.

$$\Delta t = \frac{45920J}{(365)(0.585)(5)} = \frac{326°C}{135°C} = 77 - 35°C + 35°C$$

 89,290 J of energy are removed from a 670 g block of aluminum originally at 125°C. What will the new temperature of the aluminum be? The specific heat capacity of aluminum is 0.902 U/g°C.

$$\Delta + \frac{-892905}{(6705)(0.900360)} = \frac{-14800}{+1250} = \frac{70 - 1250}{+1250}$$

Energy And Thermochemistry Review Guide Key

Brendan G. Carr

Energy And Thermochemistry Review Guide Key:

Decoding Energy And Thermochemistry Review Guide Key: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "Energy And Thermochemistry Review Guide Key," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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