



GRADE 9 NATURAL SCIENCES JUNE TEST

SECTION A

QUESTION 1

1.1 Various options are provided as answers to the following questions. Choose the correct answer and write **ONLY** the LETTER(A-D) next to the QUESTION NUMBER (1.1.1-1.1.10) on your answer sheet, for example 1.1.11 D.

1.1.1 Which of the following can you NOT learn from a chemical formula?

- | | |
|----------------------------|-------------------------|
| A Elements in the compound | B Phase of the element |
| C Ratio of atoms | D Total number of atoms |

1.1.2 Which of the following is NOT a sub-atomic particle?

- | | |
|-------------|------------|
| A Electrons | B Protons |
| C Neutrons | D Nucleons |

1.1.3 What is the prefix for 6 atoms?

- | | |
|----------|---------|
| A Hepta- | B Hexa- |
| C Penta- | D Tri- |

1.1.4 An element has an atomic number of 30 and mass number of 65. How many nucleons will it have?

- | | |
|------|------|
| A 30 | B 35 |
| C 65 | D 95 |

1.1.5 Element at group 15, period 2 on the periodic table...

- | | |
|-----------|------------|
| A Carbon | B Nitrogen |
| C Silicon | D Oxygen |

1.1.6 The following is an example of a semi-metal:

- | | |
|-------------|-----------|
| A Silicon | B Lithium |
| C Magnesium | D Carbon |

1.1.7 Substances on the left side of a chemical equation:

- | | |
|------------|--------------|
| A Products | B Reactants |
| C Metals | D Non-metals |

1.1.8 Which of the following is NOT a way to prevent rust?

- | | |
|-----------------|-----------------|
| A Galvanisation | B Electrolysis |
| C Paint | D Oil or grease |

1.1.9 Substance with a pH lower than 7 will be classified as:

- | | |
|---------------|------------|
| A Lemon Juice | B Neutral |
| C Acidic | D Alkaline |

1.1.10 An neutral substance would turn _____ with universal indicator.

- | | |
|--------------|----------|
| A Green | B Violet |
| C Colourless | D Red |

[10x1]

1.2 Provide the correct TERM for the following descriptions. Write **ONLY** the correct TERM next to the QUESTION NUMBER (1.2.1-1.2.7) on your answer sheet.

1.2.1 Measure of how strongly acidic or alkaline a substance is.

1.2.2 Reaction of a substance with oxygen.

1.2.3 Name for group 18 on the periodic table.

1.2.4 Process whereby compounds can be broken down using electricity.

1.2.5 H_2 , O_2 , Br_2 , I_2 .

1.2.6 Name for groups 3 to 10 on the periodic table.

1.2.7 When acids and bases react with each other.

[1x7]

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H Kauffman



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