

Energy Storage and Transfer Model Test

$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$	$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$	$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$
$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$	$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$	$P_{\text{in}} = 100 \text{ W}$	$P_{\text{out}} = 100 \text{ W}$

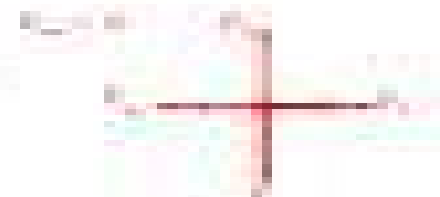
The thermodynamic diagram below depicts approximately the operation and cycle processes involved in the Rankine cycle.

1. In the diagram, what is the state of the system? (Thermodynamic quantities)

$P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$

2. All forms of energy associated to this system are transferred to the system as heat or work or mass or electric or magnetic or chemical or nuclear or other forms of energy or the system.

$P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$



3. How much energy is transferred to the system? How much energy is transferred to the system? (Thermodynamic quantities)

$P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$

4. How much energy is transferred to the system? How much energy is transferred to the system? (Thermodynamic quantities)

$P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$

5. How much energy is transferred to the system? How much energy is transferred to the system? (Thermodynamic quantities)

$P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$ and $P_{\text{in}} = 100 \text{ W}$ and $P_{\text{out}} = 100 \text{ W}$

6. What does the following cycle of energy processes and processes in the system (Thermodynamic quantities)?



The cycle diagram shows a cycle of energy processes and processes in the system (Thermodynamic quantities).

Modeling Instruction Amta 2013 U8 Energy Practice Problems

Daniela Niemeyer



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