

Stoichiometry Worksheet

I. ONE REACTANT IN EXCESS – COMBUSTION

You burn an excess of butane (C₄H₁₀) in a limited amount of oxygen (O₂). If 67.2 L of CO₂ gas is formed, what volume of oxygen gas was present at the start of the reaction?

STEP 1: Balance the reaction:



STEP 2: Use a molar conversion to convert quantity of product given to moles:

$$67.2 \text{ L CO}_2 * \frac{(1 \text{ mol CO}_2)}{(22.4 \text{ L CO}_2)} = 3 \text{ mol CO}_2$$

STEP 3: Use a molar ratio to calculate the moles of reactant that I'm asking about:

$$3 \text{ mol CO}_2 * \frac{(13 \text{ mol O}_2)}{(8 \text{ mol CO}_2)} = 4.875 \text{ mol O}_2$$

STEP 4: Use a molar conversion to convert moles of reactant to the units I ask for:

$$4.875 \text{ mol O}_2 \frac{(22.4 \text{ L O}_2)}{(1 \text{ mol O}_2)} = 109.2 \text{ L O}_2 = \boxed{109 \text{ L O}_2}$$

II. ONE REACTANT IN EXCESS – SINGLE DISPLACEMENT

14 g of lithium is combined with excess K₂SO₄. What is the mass of the new ionic compound that is formed? How do you know that this reaction will take place?

STEP 1: Balance the reaction:



Reaction occurs because Li is higher up than K on the activity series.

STEP 2: Use a molar conversion to convert quantity of reactant given to moles:

$$14 \text{ g Li} * \frac{(1 \text{ mol Li})}{(7 \text{ g Li})} = 2 \text{ mol Li}$$

STEP 3: Use a molar ratio to calculate moles of product that I'm asking about:

$$2 \text{ mol Li} * \frac{(1 \text{ mol Li}_2\text{SO}_4)}{(2 \text{ mol Li})} = 1 \text{ mol Li}_2\text{SO}_4$$

STEP 4: Use a molar conversion to convert moles of product to units that I ask for:

$$1 \text{ mol Li}_2\text{SO}_4 * \frac{(110 \text{ g Li}_2\text{SO}_4)}{(1 \text{ mol Li}_2\text{SO}_4)} = \boxed{110 \text{ Li}_2\text{SO}_4}$$

Mixed Stoichiometry Problems 2 And Answer

Kaplan Test Prep



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises [OAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests](#) Kaplan Test Prep, 2016-10-04 Kaplan's OAT 2017 2018 Strategies Practice Review provides the content review test taking strategies and realistic practice you need to get the OAT results you want Updated for the latest test changes OAT 2017 2018 is your guide to facing Test Day with confidence The Best Review Two full length online practice tests More than 600 practice questions for every subject with detailed answers and explanations 16 page tear out full color study sheets for quick review on the go A guide to the current OAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the OAT Biology General Chemistry Organic Chemistry Reading Comprehension Physics Quantitative Reasoning Kaplan's proven strategies for Test Day success Expert Guidance Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test We invented test prep Kaplan www.kaptest.com has been helping students for almost 80 years Our proven strategies have helped legions of students achieve their dreams [DAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests](#) Kaplan Test Prep, 2016-10-04 2 full length online practice tests Cover **Student Study Guide to Accompany Petrucci's General Chemistry** Robert K. Wismer, 1985 *Student Study Guide to Accompany Petrucci's General Chemistry, 3rd. Ed* Robert K. Wismer, 1982 [Kaplan PCAT 2016-2017 Strategies, Practice, and Review with 2 Practice Tests](#), 2016-02-02 Includes access to 2 full length practice tests online and detachable study sheets at the back of the book [AP Chemistry Premium, 2024: 6 Practice Tests + Comprehensive Review + Online Practice](#) Neil D. Jespersen, Pamela Kerrigan, 2023-07-04 A guide to taking the Advanced Placement exam in chemistry featuring a review of major chemistry concepts practice and diagnostic tests test taking strategies an overview of the test and practice problems [Intermetallic and Ceramic Coatings](#) Narnedra B. Dahotre, T.S. Sudarshan, 1999-02-16 Detailing the properties of specific coatings problems related to adhesion onto various substrates and potential commercial applications this text surveys up to date techniques involved in preparing intermetallic and ceramic coatings The book features a list of selected applications covering the latest industrially available practices [AP Chemistry Premium, 2026: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice](#) Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan, 2025-07 Be prepared for exam day with Barron's Trusted content from AP experts Barron's AP Chemistry Premium 2026 includes in depth content review and practice It's the only book you'll need to be prepared for exam day Written by Experienced Educators Learn from Barron's all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent changes made to the course and exam by the College Board for 2025 and beyond Get a leg up with tips strategies and study advice for exam day it's like having a trusted tutor by your side Be

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Table of Contents Mixed Stoichiometry Problems 2 And Answer

1. Understanding the eBook Mixed Stoichiometry Problems 2 And Answer
 - The Rise of Digital Reading Mixed Stoichiometry Problems 2 And Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Mixed Stoichiometry Problems 2 And Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mixed Stoichiometry Problems 2 And Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mixed Stoichiometry Problems 2 And Answer
 - Personalized Recommendations

- Mixed Stoichiometry Problems 2 And Answer User Reviews and Ratings
- Mixed Stoichiometry Problems 2 And Answer and Bestseller Lists
- 5. Accessing Mixed Stoichiometry Problems 2 And Answer Free and Paid eBooks
 - Mixed Stoichiometry Problems 2 And Answer Public Domain eBooks
 - Mixed Stoichiometry Problems 2 And Answer eBook Subscription Services
 - Mixed Stoichiometry Problems 2 And Answer Budget-Friendly Options
- 6. Navigating Mixed Stoichiometry Problems 2 And Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Mixed Stoichiometry Problems 2 And Answer Compatibility with Devices
 - Mixed Stoichiometry Problems 2 And Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mixed Stoichiometry Problems 2 And Answer
 - Highlighting and Note-Taking Mixed Stoichiometry Problems 2 And Answer
 - Interactive Elements Mixed Stoichiometry Problems 2 And Answer
- 8. Staying Engaged with Mixed Stoichiometry Problems 2 And Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mixed Stoichiometry Problems 2 And Answer
- 9. Balancing eBooks and Physical Books Mixed Stoichiometry Problems 2 And Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mixed Stoichiometry Problems 2 And Answer
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mixed Stoichiometry Problems 2 And Answer
 - Setting Reading Goals Mixed Stoichiometry Problems 2 And Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mixed Stoichiometry Problems 2 And Answer
 - Fact-Checking eBook Content of Mixed Stoichiometry Problems 2 And Answer

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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