Name:

Honors Chem Test: Ch. 7

		Choice—1 point each e choice that best completes the statemen	et or answer	s the question	
	1.	How many atoms of fluorine are present in a molecule of carbon tetrafluonde, CF.7			
		a. 1	e.	4	
		b. 2	d.	5	
	2.	Name the compound Al ₂ S ₃			
		a aluminum sulfate	· C	altuminum(II) sulfate	
		b. aluminum sulfur	di.	aluminum sulfide	
	3.	Name the compound CF.			
		a. cal crum fluoride	c.	carbon tetrafluonide	
		b. carbon fluoride	d.	mono-calcium quadraffuoride	
	4.	Name the compound N_2O_3			
		a. dimitrogen omde	C	natri e con de	
		 nitrogen triconde 	d.	dimitrogen triconde	
	51	What is the formula for diasrsenic trist	Afide?		
		a. Ar ₂ S ₂	C.	As ₂ S ₄	
		b. ArSı	d.	As ₃ S ₃	
	6.	What is the formula for sulfur dichloride?			
		a SCI	(C)	S ₂ C1	
		b SCI;	d.	S ₂ Cl ₂	
	7.	What is the formula for hydrochloric a	rcid?		
		a. HF	C.	HCIO	
		ь нсі	d.	H ₂ CO ₂	
	8. The molar mass of an element is the mass of one				
		a atom of the element.	c.	gram of the element.	
		 b. crystal of the element. 	d.	mole of the element.	
	9.	The molar mass of MgI ₂ is			
		a. the sum of the masses of 1 mol of Mg and 2 mol of I.			
		 the sum of the masses of 1 mol of Mg and 1 mol of I. 			
		c. the sum of the masses of 1 atom of Mg and 2 atoms of I.			
		d. the sum of the masses of 1 atom of Mg and 1 atom of I.			

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Robert R. Redfield

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