

NEB CLASS 11
MODEL QUESTION
(EFFECTIVE FROM 2077)

Grade XI

Subject Code : Che. 301

Time :3 hours

Attempt all questions

Subject: Chemistry (Theory)

Full marks: 75

Group 'A'

Circle the best alternative to the following questions.

(11×1 = 11)

1. How many atoms are there in two molecules of water?
 - a. 3
 - b. 4
 - c. 5
 - d. 6
2. What is the number of moles of ammonia gas formed when 0.5 mole of nitrogen gas is reacted with excess of hydrogen gas?
 - a. 0.5
 - b. 1
 - c. 2
 - d. 3
3. Which of the following bonding is responsible for the solubility of ammonia gas in water?
 - a. Hydrogen bonding
 - b. Ionic bonding
 - c. Covalent bonding
 - d. Van der Waals' force
4. What happens when Sulphur dioxide (SO_2) gas is passed through an acidified solution of hydrogen sulfide (H_2S) gas?
 - a. SO_2 is oxidized to Sulphur
 - b. H_2S is reduced to Sulphur
 - c. SO_2 is oxidized to H_2SO_4
 - d. SO_2 is reduced to Sulphur

Model Question Paper

M Walker



Model Question Paper:

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