Types of Chemical Reactions

Do atoms rearrange in predictable patterns during chemical reactions?

Why?

Recognizing patterns allows us to predict future behavior. Weather experts use patterns to predict dangerous storms so people can get their families to safety. Political analysts use patterns to predict election outcomes. Similarly, chemists classify chemical equations according to their patterns to help predict products of unknown but similar chemical reactions.

See B.

Model 1 - Types of Reactions

Set A
$4\text{Fe}(s) + 3\text{O}_3(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$
$N_y(g) + 3H_y(g) \rightarrow 2NH_y(g)$
$2SO_2(g) + O_2(g) \rightarrow 2SO_3(g)$
$MgO(s)+H_2O(l) \rightarrow Mg(OH)_3(aq)$
$P_2O_5(g) + 3H_2O(l) \rightarrow 2H_3PO_4(aq)$
$SO_{3}(g) * H_{2}O(l) \rightarrow H_{2}SO_{4}(aq)$

White the second	
$MgCO_{3}(s) \rightarrow MgO(s) + CO_{2}(g)$	
$8\text{Li}_2S(s) \rightarrow 16\text{Li}(s) + S_a(s)$	
$2H_2O(1) \rightarrow 2H_3(g) + O_2(g)$	
$2KClO_3(s) \rightarrow 2KCl(s) + 3O_2(g)$	

 $2Na_sO_s(s) \rightarrow 2Na_sO(s) + O_s(g)$

Na,CO,(aq) + CaCl,(aq) --

"
$$CaCO_3(s) + 2NaCl(aq)$$

 $FeS(s) + 2HCl(aq) \rightarrow H_2S(g) + FeCl_3(aq)$
 $HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$
 $FeBr_2(aq) + K_2PO_2(aq) \rightarrow FePO_2(s) + 3KBr(aq)$

The chemical equations in Model 1 contain the phase notations (s), (f), (g), and (aq). March
each symbol with its meaning.

dissolved in water liquid solid gas
(aq) (L) (S) (9

 Based on the examples provided, which set(s) of reactions in Model 1 typically involve ions in solution (A, B, C or D)?

C aD

 Based on the examples provided, which set(s) of reactions in Model 1 typically involve gases and/or solids?

13

Model 1 Types Of Chemical Reactions Pogil

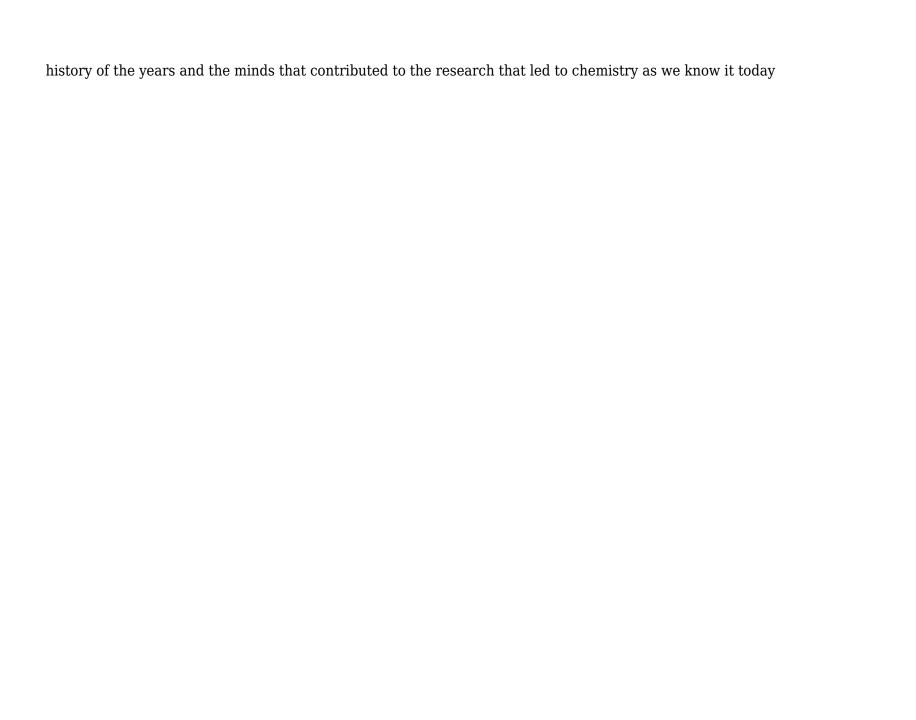
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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes Problems and Problem Solving in Chemistry Education Georgios Tsaparlis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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