

Rajiv Gandhi University of Health Sciences, Karnataka

10. Describe the extent, gross features and relations of the hypothalamus. Explain the functional importance of the hypothalamic-hypophyseal portal system. (1+1+2+1)
11. Describe the development of the umbilical cord. Explain why the umbilical arteries carry deoxygenated blood and umbilical veins carry oxygenated blood. (4+1)
12. Compare and contrast the light microscopic features of serous and mucous acini.

SHORT ANSWERS

10 x 3 = 30 Marks

13. A 30-year-old lady had to get blood drawn for some routine blood tests. At the hospital lab, the technician selected a vein that ran across the roof of the cubital fossa for the collection of blood.
 - a. Which vein did the technician select for drawing blood?
 - b. Explain the anatomical basis for the selection of this vein. (1+2)
14. Describe the location, formation and termination of the coronary sinus. (1+1+1)
15. Describe the lymphatic drainage of the lungs.
16. Name six branches of the external carotid artery.
17. Name the bones that contribute to the lateral wall of the nasal cavity. Explain the functional importance of the turbinates. (2+1)
- ~~18. Describe the attachments and extensions of the pre-vertebral fascia. (2+1)~~
19. Describe the anatomical basis for performing a lumbar puncture at the L3/L4 interspinous level.
20. Describe the three types of anastomoses with suitable examples. (1.5+1.5)
21. Describe the components of the blood-brain barrier. Explain its functional importance. (1.5+1.5)
22. Draw a neat, labelled diagram of a blastocyst.

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Michael Seilmaier



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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others

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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic

formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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web aug 23 2019 these are called polyatomic ions table 2 lists the formulas charges and names of some common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions

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web polyatomic ions are charged groups of atoms an example is ammonium ion NH_4^+ it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is NH_4Cl and ammonium sulfide is NH_4S

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web polyatomic ions have defined formulas names and charges that cannot be modified in any way table 3 4 1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example NO_3^- is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion NO_3^- contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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polyatomic ion don t add a number to the name

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web Aug 11 2022 to identify and name polyatomic ions some ions consist of groups of atoms covalently bonded together and have an overall electric charge because these ions contain more than one atom they are called polyatomic ions the structures names and formulas of some polyatomic ions are found in the below figure and table

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written