

## INORGANIC CHEMISTRY

451. Nitrogen can be obtained from air by removing  
 (a) Oxygen (b) Hydrogen (c) Carbon dioxide (d) Both (a) and (c)
452. Which of the following carbide is a methanide?  
 (a)  $\text{Be}_2\text{C}$  (b)  $\text{CaC}_2$  (c)  $\text{Al}_4\text{C}_3$  (d)  $\text{Mg}_2\text{C}_3$
453. Caro's acid is  
 (a)  $\text{H}_2\text{SO}_3$  (b)  $\text{H}_2\text{S}_2\text{O}_5$  (c)  $\text{H}_2\text{SO}_4$  (d)  $\text{H}_2\text{S}_2\text{O}_8$
454. Inorganic benzene is  
 (a)  $\text{B}_3\text{H}_3\text{N}_3$  (b)  $3\text{BH}_3\text{NH}_3$  (c)  $-\text{B}_3\text{H}_6\text{N}_3-$  (d)  $\text{H}_3\text{B}_3\text{N}_3$
455. For the reaction,  $\text{P(White)} \longrightarrow \text{P(Red)}$ ;  $\Delta H = -18.4 \text{ kJ}$ . It follows that  
 (a) Red P is readily formed from white P  
 (b) White P is readily formed from red P  
 (c) White P can not be converted to red P  
 (d) White P can be converted into red P and red P is more stable
456. Nitrogen dioxide cannot be obtained by heating  
 (a)  $\text{KNO}_3$  (b)  $\text{Pb(NO}_3)_2$  (c)  $\text{Cu(NO}_3)_2$  (d)  $\text{AgNO}_3$
457. Reactivity of borazole is greater than that of benzene because:  
 (a) Borazole is a non-polar compound  
 (b) Borazole is a polar compound  
 (c) Borazole is an electron deficient compound  
 (d) Of localized electrons in it
458. When orthophosphoric acid is heated to  $600^\circ\text{C}$ , the product formed is  
 (a) Phosphine,  $\text{PH}_3$  (b) Phosphorus pentoxide,  $\text{P}_2\text{O}_5$   
 (c) Phosphorus acid,  $\text{H}_3\text{PO}_3$  (d) Metaphosphoric acid,  $\text{HPO}_3$
459.  $\text{P}_4 + 3\text{NaOH} + 3\text{H}_2\text{O} \rightarrow \text{A} + 3\text{NaH}_2\text{PO}_2$ , here, 'A' is  
 (a)  $\text{NH}_3$  (b)  $\text{PH}_3$  (c)  $\text{H}_2\text{PO}_4$  (d)  $\text{H}_3\text{PO}_4$
460.  $\text{SO}_2$  in the solid phase exists as  
 (a) linear chain or cyclic trimer (b) planar triangular unit  
 (c) square planar unit (d) none of these
461. Sulphur does not exist as  $\text{S}_2$  molecule because  
 (a) It is less electronegative (b) Does not form  $\text{pi-pi}$  bond  
 (c) It has ability to exhibiting catenation (d) Has tendency to show variable oxidation state
462. Which of the following has S-S BOND?  
 (a)  $\text{H}_2\text{S}_2\text{O}_8$  (b)  $\text{H}_2\text{S}_2\text{O}_7$  (c)  $\text{H}_2\text{S}_2\text{O}_6$  (d) Mustard gas
463. Which of the following is correct order of stability order?  
 (a)  $\text{Pb}^{2+} > \text{Pb}^{4+}$  (b)  $\text{Pb}^{4+} > \text{Pb}^{2+}$  (c)  $\text{Si}^{2+} > \text{Si}^{4+}$  (d)  $\text{Sn}^{4+} > \text{Sn}^{2+}$
464. Quartz is a crystalline variety of  
 (a) Si (b)  $\text{SiO}_2$  (c)  $\text{Na}_2\text{SiO}_3$  (d) SiC
465. The equivalent weight of phosphoric acid in the reaction  
 $\text{NaOH} + \text{H}_3\text{PO}_4 \longrightarrow \text{NaH}_2\text{PO}_4 + \text{H}_2\text{O}$   
 (a) 25 (b) 49 (c) 59 (d) 98

# Inorganic Chemistry Objective Type Questn

**Michael Brown**



**Inorganic Chemistry Objective Type Questn:**

## Unveiling the Magic of Words: A Overview of "**Inorganic Chemistry Objective Type Questn**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Inorganic Chemistry Objective Type Questn**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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