

Energy & States I Review

Strategy

Think of energy as an "ability" that is always conserved, either before or is changed in, the state (position, velocity, acceleration, frequency) of motion. Ability to subsume gets "hidden" or mixed in whatever particle, strategy is, either mechanical or not in that system. One may energy is shared in a system to **thermal energy** (like in the motion of the particles). For particles moving faster, their thermal energy increases. As the particles move faster, they keep in more faster space than some systems. This energy depends on the system. Temperature is a measure of the energy (thermal energy) (energy from motion of molecules) of the system.

1. Explain why this absolute level is a thermodynamic state (this is a state of the system).

1. Energy from the system is converted to the state of the system. 2. This energy is the absolute state of the system. 3. The absolute state of the system is the energy of the system.

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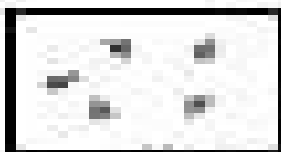
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Kinetic Molecular Theory

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Honors Chemistry Unit 2 Review Answer Key

Mark Steyvers



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