# Concept Review:

- 1. Draw the Lewis diagram for each compound.
  - A. a molecule composed of two chlorine atoms
  - B. a molecule composed of a hydrogen atom and a bromine atom

## Answers:

### Solution

Chlorine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of two chlorine atoms is similar to the one for fluorine:

:ci:ci:

Bromine has the same valence shell electron configuration as fluorine, so the Lewis diagram for a molecule composed of a hydrogen atom and a bromine atom is similar to that for hydrogen and fluorine:

H:Br:

# Test Yourself:

- 1. Draw the Lewis diagram for each compound.
  - A. a molecule composed of one chlorine atom and one fluorine atom
  - B. a molecule composed of one hydrogen atom and one iodine atom

# **Holt Chemistry Concept Review Covalent Bonds Answers**

**Ian Pickup** 

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