# Hydrocarbons

| Methane                                | General formula for an alkane               |  | General formula for an alkene   | Ethene   | Propene                         | Butene   |  |
|--|---|--|---|--|---------------------------------|--|--|
|  | How to test between<br>an alkene and alkane |  | Alkane or alkene?  C/H <sub>4</sub> C/H <sub>6</sub> C/H <sub>6</sub> C/H <sub>6</sub> C/H <sub>6</sub> |  |                                 |  |  |
| Ethane                                 |   |  |   | Show how ethere is polymerised   |                                 |  |  |
|  | Definition of a hydrocarbon                 |  |   |  |                                 |  |  |
| Propune                                |   | What is meant by saturated/unsaturated |   | Draw the monomer that makes this polymer  Br H Br H Br H  I I I I I  H-C-C-C-C-C-C-H  I I I I I  H H H H H H |                                 |  |  |
| Butane                                 |   | How ethanol is produced<br>from ethane |   | Describe the proce<br>Process:   | ns of cracking                  | Complete the equations for cracking  C <sub>0</sub> H <sub>10</sub> → C <sub>1</sub> H <sub>4</sub> + + C <sub>4</sub> H <sub>30</sub> |  |
| A way to remember the order of alkanes |   | _by fermentation                       |   | Types of products.  Conditions:  Why it is done:   | The types of products produced: |  |  |

# **Hydrocarbon Compounds Core Teaching Resources**

**Rachel S Tattersall** 

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