

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

Use only the student's concordant results

mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol} \quad \checkmark$$

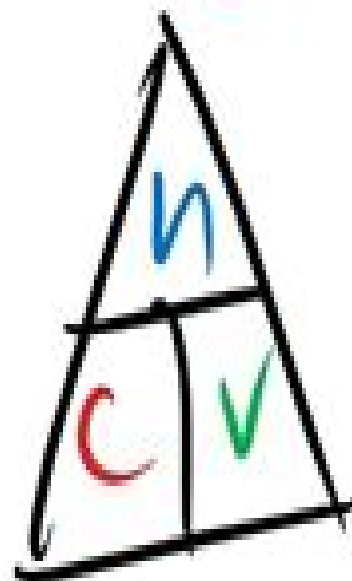
$$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol} \quad \checkmark$$

$$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3 \quad \checkmark$$

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



GCSE CHEMISTRY

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LM Reder



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Essential Skills for GCSE Chemistry Nora Henry,2019-07-29 Build essential maths literacy and working scientifically skills to boost marks in GCSE Chemistry and ensure that students reach their full potential Suitable for all specifications this skills book provides additional support and will help to Sharpen mathematical skills with plenty of practice questions and coverage of all the maths techniques needed for the exams Improve literacy skills with tips on how to write longer answers plus peer assessment marking activities Develop the working scientifically skills needed to plan carry out and evaluate practical experiments in order to secure the maximum number of marks Build confidence by putting skills into practice using our three step formula students will progress from worked examples to guided questions and exam style questions with fully worked solutions in the book Raise performance in the exams with practical advice on how to revise effectively and tips on understanding the questions command words and assessment objectives

Gcse Success Rev Gd Aqa Chem Christine Horbury,2008 This Success Revision Guide offers accessible content to help students manage their revision and prepare for the exam efficiently The content is broken into manageable sections and advice is offered to help build students confidence Exam tips and techniques are provided to support students throughout the revision process

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