

Wintersemester 2019/2020 - Klausuraufgaben 8B

Aufgabe 8 (max. 70 Punkte/100% Gesamtpunkte)

Die nachstehenden Probleme sollen Sie zur Vertiefung des wissenschaftlichen Verständnisses der Stoffeigenschaften (Kapitel 10) befähigen. Sie sind, soweit in der Aufgabenstellung nicht anders angegeben, unter folgenden Bedingungen zu lösen:

$$p = p_0 = 1 \text{ bar} = 10^5 \text{ Pa}$$

1. Chlorwasserstoff (molekulare Masse $M = 36,5 \text{ g/mol}$) wird bei einer Temperatur von 25°C verdichtet. Wie groß ist die Dichte des resultierenden Gases?

$$\rho = 1,97 \text{ g/L} \quad \text{[Literaturwert: } \rho = 1,88 \text{ g/L]} \\ \text{[Dichte des Gases bei } 25^\circ\text{C und } p_0]$$

2. Ethylenglykol (molekulare Masse $M = 62 \text{ g/mol}$) wird bei 25°C verdichtet. Wie groß ist die Dichte des resultierenden Gases?

$$\rho = 1,25 \text{ g/L} \quad \text{[Literaturwert: } \rho = 1,25 \text{ g/L]} \\ \text{[Dichte des Gases bei } 25^\circ\text{C und } p_0]$$

3. Ethanol ist ein verdampfungsfähiges flüchtiges Gas. Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^5 Pa ? Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^6 Pa ?

$$\rho = 1,25 \text{ g/L} \quad \text{[Literaturwert: } \rho = 1,25 \text{ g/L]} \\ \text{[Dichte des Gases bei } 25^\circ\text{C und } p_0]$$

4. In der folgenden Tabelle ist die Dichte des flüchtigen Gases bei 25°C angegeben. Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^5 Pa ? Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^6 Pa ?

$$\rho = 1,25 \text{ g/L} \quad \text{[Literaturwert: } \rho = 1,25 \text{ g/L]} \\ \text{[Dichte des Gases bei } 25^\circ\text{C und } p_0]$$

5. Ethanol ist ein verdampfungsfähiges flüchtiges Gas. Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^5 Pa ? Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^6 Pa ?

$$\rho = 1,25 \text{ g/L} \quad \text{[Literaturwert: } \rho = 1,25 \text{ g/L]} \\ \text{[Dichte des Gases bei } 25^\circ\text{C und } p_0]$$

6. Ethanol ist ein verdampfungsfähiges flüchtiges Gas. Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^5 Pa ? Wie groß ist die Dichte des resultierenden Gases bei 25°C und einer Verdichtungsdruck von 10^6 Pa ?

Gases Section 3 Review Answers

Robert R. Redfield



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