



**Western
Science**

Chemistry 1302B Final Exam *Answer Key*

Wednesday, April 19, 2023
2:00 – 5:00 pm

1. The density of the $\text{N}_2(\text{g})$ in a rigid tank at 250 kPa was 1.54 g L^{-1} . What was the temperature of the gas in this container?

547 K

2. Which of the following statements is true regarding the two samples below?

- Sample I: 3.0 moles of Ar at 20 °C in a container of volume V
- Sample II: 3.0 moles of Ar at 100 °C in a container of volume V

The pressure of Sample II is greater than that of Sample I.

3. Before contact with Europeans, Indigenous Peoples in this region produced maple syrup by evaporating the water from maple sap held in wooden containers. To heat the sap, they would drop in some stones that had been heated in a fire. If a 500 g stone that had been heated to 200 °C was added to a container holding 2.00 kg of sap at 5 °C, what temperature would the liquid have achieved when stone and sap reached thermal equilibrium? (Helpful data: The specific heat capacity of the stone was $0.790 \text{ J g}^{-1} \text{ K}^{-1}$ and that of the sap at this stage was the same as that of liquid water. Ignore heat losses to the container and surroundings.)

53.8 °C

4. Given: $\Delta H_f^\circ(\text{CO}_2\text{g}) = -130 \text{ kJ mol}^{-1}$, the ΔH° of the reaction $\text{CO}(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{CO}_2(\text{g})$ is:

- I. ΔH_f° of $\text{CO}_2(\text{g})$
- II. The heat of combustion of $\text{CO}(\text{g})$
- III. -504 kJ

II only

5. Given: $\Delta H_{\text{sub}}^\circ = 402.6 \text{ kJ mol}^{-1}$ $\Delta H_{\text{fus}}^\circ = 535.6 \text{ kJ mol}^{-1}$

Estimate ΔH for the reaction: $\text{BF}_3(\text{g}) + \text{BCl}_3(\text{g}) \rightarrow \text{BF}_3\text{Cl}_2(\text{g}) + \text{BF}_2\text{Cl}_2(\text{g})$

0.0 kJ

6. How many of the processes below have a positive ΔS° ?

- $\text{N}_2(\text{g}) + 3 \text{ H}_2(\text{g}) \rightarrow 2 \text{ NH}_3(\text{g})$
- $2 \text{ NaCl}(\text{s}) + \text{Cl}_2(\text{g}) \rightarrow 2 \text{ NaCl}(\text{s})$
- Heating 10 g of water from 0 °C to 10 °C
- Condensing 10 g of water vapour into liquid at 100 °C

Gases Section 1 Answers

D Kirk

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located on the right side of the slide, partially overlapping the light blue bar.

Gases Section 1 Answers:

Solutions Manual to Accompany Elements of Physical Chemistry David Smith, 2013-05-30 The Solutions Manual to accompany Elements of Physical Chemistry 6th edition contains full worked solutions to all end of chapter discussion questions and exercises featured in the book The manual provides helpful comments and friendly advice to aid understanding It is also a valuable resource for any lecturer who wishes to use the extensive selection of exercises featured in the text to support either formative or summative assessment and wants labour saving ready access to the full solutions to these questions Worked Problems in Heat, Thermodynamics and Kinetic Theory for Physics Students L. Pincherle, 2013-10-22 Worked Problems in Heat Thermodynamics and Kinetic Theory for Physics Students is a complementary to textbooks in physics This book is a collection of exercise problems that have been part of tutorial classes in heat and thermodynamics at the University of London This collection of exercise problems with answers that are fully worked out deals with various topics This book poses problems covering the definition of temperature such as calculating the assigned value of the temperature of boiling water under specific conditions This text also gives example of problems dealing with the first law of thermodynamics and with the definition of thermal capacities Some practical questions such as problems dealing with thermal engines are presented This book then discusses problems using the energy equation as well as asking the student to derive a general equation of state of a material satisfying a specific condition This text challenges the student to use a T S diagram to calculate the efficiency of a reversible cycle under certain conditions Several other problems concern the Joule and Joule Kelvin effects low temperature physics and heat conduction This review material can be helpful for students of physics thermodynamics and related subjects It can also be used by teachers of physics *Oswaal One For All Olympiad Previous Years' Solved Papers Class 3 (Set of 6 Books) Maths, English, Science, Reasoning, Cyber & General Knowledge (For 2023 Exam)* Oswaal Editorial Board, 2023-06-14 Description of the product Crisp Revision with Concept wise Revision Notes Mind Maps 100% Exam Readiness with Previous Years Questions 2011 2022 from all leading Olympiads like IMO NSO ISO Hindustan Olympiad Valuable Exam Insights with 3 Levels of Questions Level 1 2 Achievers Concept Clarity with 500 Concepts 50 Concepts Videos Extensive Practice with Level 1 Level 2 Practice Papers *Report of the ... Meeting of the British Association for the Advancement of Science* British Association for the Advancement of Science, 1888 *The Coal Trade Bulletin*, 1921 *The Atmospheric Environment* Michael B. McElroy, 2021-08-10 This comprehensive introduction to the physics and chemistry of Earth's atmosphere explains the science behind some of the most critical and intensely debated environmental controversies of our day In it one of the world's leading experts on planetary environments presents the background necessary to assess the complex effects of human activity on our atmosphere and climate Unique in its breadth and depth of coverage The Atmospheric Environment includes a survey of Earth's climatic history to provide a context for assessing the changes underway today It is written for and will be of lasting value to a varied audience including not only

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QBase Anaesthesia with CD-ROM: Volume 6, MCQ Companion to Fundamentals of Anaesthesia Colin Pinnock, Robert Jones, Simon Maguire, Julian M. Barker, Simon Mills, 2004-03-18 Another title in GMM's highly successful QBase series this time focusing on the Primary FRCA Fellowship of the Royal College of Anaesthetists with sets of examination papers the candidate can use to revise and practice their MCQ technique As always the self learning experience is greatly enhanced by the free CD ROM containing the QBase examination software This powerful program allows the user to sit the pre set exams as printed in the book or to create their own exams using questions drawn from the total pool available on the CD ROM The program can either generate these exams randomly maintaining the same proportions of each subject as the pre set exams or the user can select any number of questions in any subject area to create their very own exam However the user chooses to set an exam they can then mark analyse and store each attempt and review and re sit the same exam at a later date comparing their scores with previous attempts When re sitting an exam the user can also choose to shuffle the question stems A E so that they cannot simply remember patterns of true and false answers A final unique aspect of the program is that it allows the user to select how confident they are of their answer and the program can then provide feedback on a user's guessing strategy

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