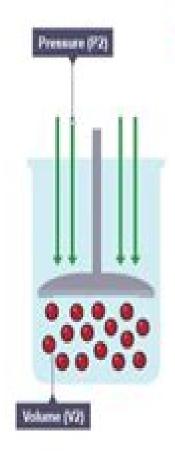
Volume and pressure in gases - the gas laws Boyle's law





Decreasing the volume of a gas increases the pressure of the gas. An example of this is when a gas is trapped in a cylinder by a piston. If the piston is pushed in, the gas particles will have less room to move as the volume the gas occupies has been decreased.

Gases And Pressure Section 1 Review Answers

John N. McCarty

Gases And Pressure Section 1 Review Answers:

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