

Gas Laws and Theory Quiz SCH3U

Name: _____

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(Product) 7 7 4 6

Assessment of Learning

$P = 8.31 \text{ kPa} \cdot \text{L} / \text{mol} \cdot \text{K} \times 1 \text{ mol} = 101.325 \text{ kPa} = 760 \text{ torr} = 760 \text{ mmHg}$ SATP = 25°C and 100 kPa
STP = 0°C and 100.3 kPa

Multiple Choice. Circle the most correct answer. (780)

1. A pressure reading of 103.4 kPa is equivalent to _____ mmHg.
A) 653.4 B) 775.6 C) 774.6 D) 782.4

2. The particles that make up a solid move _____ than do the particles that make up a gas.
A) more quickly and farther C) more quickly
B) more slowly D) in the same way

3. According to the kinetic-molecular theory, how does a gas expand?
A) Its particles become larger. B) Collisions between particles become elastic.
C) Its temperature decreases. D) Its particles move greater distances apart.

4. If the temperature of a container of gas remains constant, how could the pressure of the gas increase?
A) The mass of the gas molecules increase B) The diffusion of the gas molecules increases.
C) The size of the container increases D) The number of gas molecules in the container increases.

5. Which of the following laws shows an inverse relationship between pressure and volume?
A) Boyle's Law B) Charles's Law C) Graham's Law D) Dalton's Law

6. A gaseous mixture at a total pressure of 1.50 atm contains equal molar amounts of He, Ne, and Ar. At constant temperature CO_2 gas is added to the mixture until the total pressure is 3.00 atm. Which of the following is a correct statement concerning partial pressures after the CO_2 addition?
A) The partial pressure of Ar has doubled.
B) The partial pressure of CO_2 is 3 times that of Ne.

Gas Laws Quiz Answers

Sheri Berg



Gas Laws Quiz Answers:

Chemistry All-in-One For Dummies (+ Chapter Quizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test Class 11-12 Chemistry MCQ (Multiple Choice Questions) Arshad Iqbal, 2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model

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400 online questions arranged by topic for customized practice online practice includes answer explanations with expert advice for all questions plus scoring to track your progress This essential guide is the perfect practice supplement for students and teachers

Engineering Thermodynamics Dr. S. S. Khandare, 2003 This book on Engineering Thermodynamic contains basic principles and fundamental laws of Thermal Engineering It deals with the gas laws and properties of fluids like pressure temperature and volume The book discusses the thermodynamic processes like isothermal isentropic and polytropic processes The new concept of availability and irreversibility has been included in the book The various properties like enthalpy entropy internal energy of steam are discussed The topics on properties of steam and steam cycles like rankine modified rankine cycles are also presented in the book

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follow text which emphasizes the essential coherence of the material Anesthesia Review: 1000 Questions and Answers to Blast the BASICS and Ace the ADVANCED Sheri Berg,2018-06-04 With contributors from Massachusetts General Hospital and Harvard Medical School the unique and thorough Anesthesia Review 1000 Questions and Answers to Blast the BASICS and Ace the ADVANCED covers both BASIC or ADVANCED levels of Anesthesiology training in a single volume Any resident in Anesthesiology will find a gold mine of material including topic specific chapters with exam like questions answers with explanations and references for further in depth review for fast efficient preparation **Oswal - Gurukul Chemistry Most**

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Engineers Jack Hipple,2017-02-06 Outlines the concepts of chemical engineering so that non chemical engineers can interface with and understand basic chemical engineering concepts Overviews the difference between laboratory and industrial scale practice of chemistry consequences of mistakes and approaches needed to scale a lab reaction process to an operating scale Covers basics of chemical reaction engineering mass energy and fluid energy balances how economics are scaled and the nature of various types of flow sheets and how they are developed vs time of a project Details the basics of fluid flow and transport how fluid flow is characterized and explains the difference between positive displacement and centrifugal pumps along with their limitations and safety aspects of these differences Reviews the importance and approaches to controlling chemical processes and the safety aspects of controlling chemical processes Reviews the important chemical engineering design aspects of unit operations including distillation absorption and stripping adsorption evaporation and crystallization drying and solids handling polymer manufacture and the basics of tank and agitation system design

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