

vogadro's Law:

$$V=an$$
, or $\frac{V_1}{n_1}=\frac{V_2}{n_2}$

Bloyle's Law:

Charles' Law:

Dalton's Law:

IDEAL GAS LAW:

R= 0.08206 L.atn

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Conversion Factors

1 atm = 760 mm Hq

= 760 torr = 101325 Pa

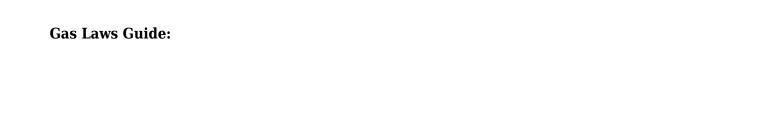
= 101.325 kPa = 1.01325 bar = 29.92 in Hg

= 14.7 lb/in2 = 101325 N/m2 = pressure at

sea level

Gas Laws Guide

Pamela Mills, Amina El-Ashmawy



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