

Identification of a Metal Carbonate

Experiment 3: A study in stoichiometry

Metal carbonates react with hydrochloric acid to produce water and carbon dioxide gas. This is the "acid test" used by geologists to identify carbonate containing rocks and minerals. The equation for this reaction, as it pertains to this experiment, is:



or for a bicarbonate (hydrogen carbonate):



Carbon dioxide is fairly soluble in water under high pressure conditions, as evidenced in carbonated soft drinks, however at normal atmospheric pressure the carbon dioxide escapes and the carbonated beverage goes flat if left without being tightly capped.

In this experiment, a weighted sample of unknown metal carbonate will chemically react and the mass of carbon dioxide escaping will be determined. This will provide the data needed to calculate the molar mass of the unknown and determine which of the substances listed in Table 1 can be identified as the unknown metal carbonate. **WARNING!** Some of the unknowns as well as the hydrochloric acid used in this experiment are very corrosive substances. Wear Eye Protection and avoid skin contact. Wash with running water if exposed.

PRE-LAB ASSIGNMENT: In the space provided in Table 1, write the chemical name and molar mass for each of the following carbonates.

Table 1: Metal Carbonate Unknowns

CHEMICAL NAME	CHEMICAL FORMULA	MOLAR MASS
	Na₂CO₃	
	Na₂CO₃ · ½ H₂O	
	NaHCO₃	
	K₂CO₃	
	KHCO₃	
	K₂CO₃ · ½ H₂O	

Metal Carbonate With Hydrochloric Acid Experiment Isa

Daniela Niemeyer



Metal Carbonate With Hydrochloric Acid Experiment Isa:

This book delves into Metal Carbonate With Hydrochloric Acid Experiment Isa. Metal Carbonate With Hydrochloric Acid Experiment Isa is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Metal Carbonate With Hydrochloric Acid Experiment Isa, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Metal Carbonate With Hydrochloric Acid Experiment Isa
 - Chapter 2: Essential Elements of Metal Carbonate With Hydrochloric Acid Experiment Isa
 - Chapter 3: Metal Carbonate With Hydrochloric Acid Experiment Isa in Everyday Life
 - Chapter 4: Metal Carbonate With Hydrochloric Acid Experiment Isa in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Metal Carbonate With Hydrochloric Acid Experiment Isa. This chapter will explore what Metal Carbonate With Hydrochloric Acid Experiment Isa is, why Metal Carbonate With Hydrochloric Acid Experiment Isa is vital, and how to effectively learn about Metal Carbonate With Hydrochloric Acid Experiment Isa.
 3. In chapter 2, this book will delve into the foundational concepts of Metal Carbonate With Hydrochloric Acid Experiment Isa. The second chapter will elucidate the essential principles that must be understood to grasp Metal Carbonate With Hydrochloric Acid Experiment Isa in its entirety.
 4. In chapter 3, the author will examine the practical applications of Metal Carbonate With Hydrochloric Acid Experiment Isa in daily life. The third chapter will showcase real-world examples of how Metal Carbonate With Hydrochloric Acid Experiment Isa can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Metal Carbonate With Hydrochloric Acid Experiment Isa in specific contexts. The fourth chapter will explore how Metal Carbonate With Hydrochloric Acid Experiment Isa is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Metal Carbonate With Hydrochloric Acid Experiment Isa. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Metal Carbonate With Hydrochloric Acid Experiment Isa.

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