

Lab Report 2 – Characterization of Synthesis of Cadmium

Name: **Elisa Kuhnert**

Matriculation No.: **21**

Course: **Chemie 1**

Reaction:

Effect of persulfate on the solubility of cadmium

Sample	Amount of Cadmium (mg)	Amount of Sulfate (mg)
1	0.0000	0.0000
2	0.0000	0.0000
3	0.0000	0.0000
Amount of Cd^{2+} in		0.0000
Amount of $\text{S}_2\text{O}_8^{2-}$ in		0.0000

Reaction: Pb^{2+} and Cu^{2+} are precipitated from the solution

1. Characterization of the reaction

Write the reaction of Pb^{2+} and Cu^{2+} with $\text{S}_2\text{O}_8^{2-}$ and $\text{S}_2\text{O}_8^{2-}$ in the form of a reaction

2. Characterization of the reaction

Write the reaction of Pb^{2+} and Cu^{2+} with $\text{S}_2\text{O}_8^{2-}$ in the form of a reaction

Reaction: Pb^{2+} and Cu^{2+}

Write the reaction of Pb^{2+} and Cu^{2+} with $\text{S}_2\text{O}_8^{2-}$ in the form of a reaction

Write the reaction of Pb^{2+} and Cu^{2+} with $\text{S}_2\text{O}_8^{2-}$ in the form of a reaction

Reaction: Pb^{2+} and Cu^{2+} are precipitated from the solution

Write the reaction of Pb^{2+} and Cu^{2+} with $\text{S}_2\text{O}_8^{2-}$ in the form of a reaction

Reaction: Pb^{2+} and Cu^{2+} are precipitated from the solution

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