

# Characteristics of Gases

1. Gases or their mixtures are homogeneous in composition.
2. Gases possess high diffusibility.
3. Gases can be liquified, if subjected to low temperatures (below critical) or high pressures.
4. Gases have very low density due to negligible intermolecular forces.
5. Gases do not have definite shape and volume like liquids.
6. Gases undergo similar change with the change of temperature and pressure. In other words, gases obey certain laws known as gas laws.
7. Gases have infinite expansibility and high compressibility.
8. Gases have highest kinetic energy.
9. Gases exert pressure.
10. Gaseous molecules collide with one another and also with the walls of container with perfectly elastic collisions.
11. Thermal energy of gases  $\gg$  molecular attraction.

# Gaseous State Multiple Choice Questions

**Singhal Atul**



## **Gaseous State Multiple Choice Questions:**

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