

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|------------------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **decreases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

Exploring Periodic Trends Answer Key

S Marginson



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final section examines the role of key institutions and bodies in Africa in mitigating the impact of conflict on the socioeconomic lives of the people in the region Bodies discussed include the African Union Economic Community of West African States Southern African Development Community and the East African Community The book will be an invaluable resource for researchers academics and policymakers working in the areas of law political science human rights peace studies development studies and African studies *The GEO Handbook on Biodiversity Observation Networks* Michele Walters, Robert J. Scholes, 2016-11-25 Biodiversity observation systems are almost everywhere inadequate to meet local national and international treaty obligations As a result of alarmingly rapid declines in biodiversity in the modern era there is a strong worldwide desire to upgrade our monitoring systems but little clarity on what is actually needed and how it can be assembled from the elements which are already present This book intends to provide practical guidance to broadly defined biodiversity observation networks at all scales but predominantly the national scale and higher This is a practical how to book with substantial policy relevance It will mostly be used by technical specialists with a responsibility for biodiversity monitoring to establish and refine their systems It is written at a technical level but one that is not discipline bound it should be intelligible to anyone in the broad field with a tertiary education **Prentice Hall Science Explorer** Michael J. Padilla, 2002 **Bulletin of the Atomic Scientists** , 1984-08

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