# **Balancing Chemical Equations**

Balance the equations below:

3) 
$$Z NaCl + I F_2 \rightarrow Z NaF + I Cl_2$$

4) 
$$Z H_2 + I O_2 \rightarrow Z H_2O$$

5) 
$$\underline{I}$$
 Pb(OH)<sub>2</sub> +  $\underline{Z}$  HCI  $\rightarrow$   $\underline{Z}$  H<sub>2</sub>O +  $\underline{I}$  PbCl<sub>2</sub>

8) 
$$/ C_3H_6 + 5O_2 \rightarrow 3CO_2 + 4H_2O$$

16) 
$$Z K + 1 MgBr \rightarrow Z KBr + 1 Mg$$

19) 
$$Z H_2O + I O_2 \rightarrow Z H_2O_2$$

20) 
$$2$$
 NaBr +  $2$  CaF<sub>2</sub>  $\rightarrow$   $2$  NaF +  $2$  CaBr<sub>2</sub>

21) 
$$\underline{\hspace{1cm}}$$
  $H_2SO_4 + \underline{\hspace{1cm}}$   $Z$   $NaNO_2 \rightarrow \underline{\hspace{1cm}}$   $Z$   $HNO_2 + \underline{\hspace{1cm}}$   $L$   $Na_2SO_4$ 

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