

Student Exploration: Ionic Bonds

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Nate can give Clara a marker

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give Fred and Florence one marker each

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The Ionic Bonds Gizmo™ allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium: 1

Chlorine: 7

2. Click **Pause** (⏏). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

sodium

Which is a nonmetal?

chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

The chlorine atom took the electron and added it to its outermost shell

Explore Learning Student Exploration Ionic Bonds Key

Shasha Hu



Explore Learning Student Exploration Ionic Bonds Key:

Whispering the Secrets of Language: An Psychological Journey through **Explore Learning Student Exploration Ionic Bonds Key**

In a digitally-driven world wherever monitors reign supreme and immediate transmission drowns out the subtleties of language, the profound secrets and psychological nuances hidden within words usually go unheard. However, set within the pages of **Explore Learning Student Exploration Ionic Bonds Key** a captivating literary treasure pulsing with natural thoughts, lies an exceptional quest waiting to be undertaken. Composed by a talented wordsmith, this marvelous opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within ab muscles cloth of each and every word. Within the mental depths of this poignant evaluation, we shall embark upon a honest exploration of the book is key styles, dissect its captivating writing type, and yield to the powerful resonance it evokes heavy within the recesses of readers hearts.

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