

Name: Answer Key Date: _____ Per.: _____

Covalent Bond Practice

1. Fill in the missing information on the chart.

Element	# of Protons	# of Electrons	# of Valence Electrons	# of electrons to fill outer shell.
Carbon	6	6	4	4
Hydrogen	1	1	1	1
Chlorine	17	17	7	1
Helium	2	2	2	0
Phosphorus	15	15	5	3
Oxygen	8	8	6	2
Sulfur	16	16	6	2
Nitrogen	7	7	5	3

2. For each of the following covalent bonds:

- Write the symbols for each element.
- Draw a Lewis Dot structure for the valence shell of each element.
- Rearrange the electrons to pair up electrons from each atom.
- Draw circles to show the sharing of electrons between each pair of atoms.
- Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
- Write the chemical formula for each molecule.

a) Hydrogen + Hydrogen



b) Chlorine + Chlorine



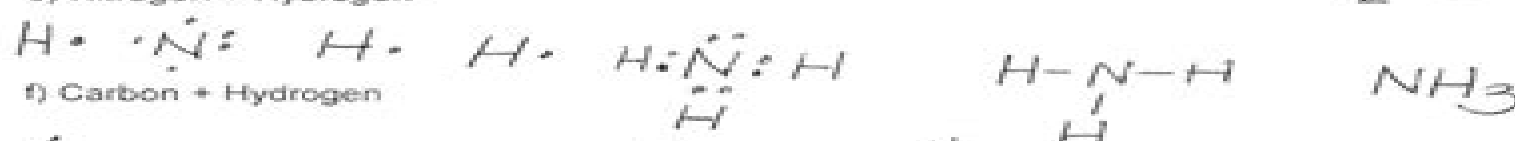
c) Hydrogen + Chlorine



d) Hydrogen + Oxygen



e) Nitrogen + Hydrogen



f) Carbon + Hydrogen



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Y Pai



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Table of Contents Explore Learning Student Exploration Covalent Bonds Answer Key

1. Understanding the eBook Explore Learning Student Exploration Covalent Bonds Answer Key
 - The Rise of Digital Reading Explore Learning Student Exploration Covalent Bonds Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Student Exploration Covalent Bonds Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Student Exploration Covalent Bonds Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Student Exploration Covalent Bonds Answer Key

- Personalized Recommendations
 - Explore Learning Student Exploration Covalent Bonds Answer Key User Reviews and Ratings
 - Explore Learning Student Exploration Covalent Bonds Answer Key and Bestseller Lists
5. Accessing Explore Learning Student Exploration Covalent Bonds Answer Key Free and Paid eBooks
- Explore Learning Student Exploration Covalent Bonds Answer Key Public Domain eBooks
 - Explore Learning Student Exploration Covalent Bonds Answer Key eBook Subscription Services
 - Explore Learning Student Exploration Covalent Bonds Answer Key Budget-Friendly Options
6. Navigating Explore Learning Student Exploration Covalent Bonds Answer Key eBook Formats
- ePub, PDF, MOBI, and More
 - Explore Learning Student Exploration Covalent Bonds Answer Key Compatibility with Devices
 - Explore Learning Student Exploration Covalent Bonds Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Explore Learning Student Exploration Covalent Bonds Answer Key
 - Highlighting and Note-Taking Explore Learning Student Exploration Covalent Bonds Answer Key
 - Interactive Elements Explore Learning Student Exploration Covalent Bonds Answer Key
8. Staying Engaged with Explore Learning Student Exploration Covalent Bonds Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Student Exploration Covalent Bonds Answer Key
9. Balancing eBooks and Physical Books Explore Learning Student Exploration Covalent Bonds Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Student Exploration Covalent Bonds Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Explore Learning Student Exploration Covalent Bonds Answer Key
- Setting Reading Goals Explore Learning Student Exploration Covalent Bonds Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explore Learning Student Exploration Covalent Bonds Answer Key

- Fact-Checking eBook Content of Explore Learning Student Exploration Covalent Bonds Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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