

Activity A:

Subatomic particles

Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons (N) in an atom?
- C. Which particle (proton, neutron, or electron) has a
Positive charge?
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge?

3. **Observe:** Turn on **Electron configuration**. An atom's electrons form an electron cloud. Electrons are arranged in shells. The number of shells is equal to the atomic number. The number of electrons in each shell is also equal to the number of protons in the atom. The number of protons is equal to the number of electrons in a neutral atom.

- A. What are the shells called?
- B. What are the shells called?
- C. What are the shells called?

Electron shells

Electron shells

Electron shells

4. **Observe:** Turn on **Electron configuration**. The following questions:

- A. How many electrons are in a neutral atom of Helium?
- B. How many electrons are in a neutral atom of Hydrogen?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 1 neutron?
- E. How many electrons are in a neutral atom of Helium?

2

1

4

3

2

Explore Learning Element Builder Gizmo Answers

JE Gale



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Table of Contents Explore Learning Element Builder Gizmo Answers

1. Understanding the eBook Explore Learning Element Builder Gizmo Answers
 - The Rise of Digital Reading Explore Learning Element Builder Gizmo Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Element Builder Gizmo Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Element Builder Gizmo Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Element Builder Gizmo Answers

- Personalized Recommendations
- Explore Learning Element Builder Gizmo Answers User Reviews and Ratings
- Explore Learning Element Builder Gizmo Answers and Bestseller Lists
- 5. Accessing Explore Learning Element Builder Gizmo Answers Free and Paid eBooks
 - Explore Learning Element Builder Gizmo Answers Public Domain eBooks
 - Explore Learning Element Builder Gizmo Answers eBook Subscription Services
 - Explore Learning Element Builder Gizmo Answers Budget-Friendly Options
- 6. Navigating Explore Learning Element Builder Gizmo Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Element Builder Gizmo Answers Compatibility with Devices
 - Explore Learning Element Builder Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Element Builder Gizmo Answers
 - Highlighting and Note-Taking Explore Learning Element Builder Gizmo Answers
 - Interactive Elements Explore Learning Element Builder Gizmo Answers
- 8. Staying Engaged with Explore Learning Element Builder Gizmo Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Element Builder Gizmo Answers
- 9. Balancing eBooks and Physical Books Explore Learning Element Builder Gizmo Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Element Builder Gizmo Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Element Builder Gizmo Answers
 - Setting Reading Goals Explore Learning Element Builder Gizmo Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Element Builder Gizmo Answers

- Fact-Checking eBook Content of Explore Learning Element Builder Gizmo Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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