

Page 2	Mark Scheme	Syllabus	Paper
	IGCSE – May/June 2008	0620	31

An incorrectly written symbol, e.g. NA or CL, should be penalised once in a question.

- 1 (a) bromine [1]
- (b) germanium [1]
- (c) potassium **or** calcium [1]
- (d) krypton [1]
- (e) iron **or** cobalt [1]
- (f) bromine [1]
- (g) vanadium [1]

ACCEPT name or symbol

[Total: 7]

2 (a)

electron	e^- or e	1/1840 or 1/2000 or 0 1/1837 or negligible	-1
proton	p or p^+ or H^+	1	+1
neutron	n	1	0 or neutral

each correct row (1)

[3]

- (b) (i) equal numbers of protons and electrons of positive and negative charges **or** charges cancel/balance [1]
or net charge = 0 [1]
- (ii) lose electron(s) [1]
more protons than electrons [1]
NOT more + than –
- (iii) different numbers of neutrons [1]
same number of protons **or** same number of electrons [1]
for just giving- they are isotopes [1] **ONLY**
- (iv) an element is known for each proton number [1]
accept any sensible idea, for example no gaps between $z = 1$ and $z = 103$

[Total: 10]

Marking Scheme Of 0620 32 M J 08

Michael Hoelscher



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