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An Industry Super Fund

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January

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30						

S Scheduled RDO
 F Flexible RDO
 L Designated Long Weekend (Public Holiday & RDO)
 S School Holidays
 A Annual Leave

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(LIFT SECTOR)

Long Service Leave Rdo Calendar 2015

Michael Seilmaier

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the name bar.

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Whispering the Strategies of Language: An Mental Quest through **Long Service Leave Rdo Calendar 2015**

In a digitally-driven world wherever monitors reign great and quick communication drowns out the subtleties of language, the profound strategies and mental nuances concealed within words usually get unheard. Yet, set within the pages of **Long Service Leave Rdo Calendar 2015** a interesting fictional value pulsing with fresh emotions, lies a fantastic journey waiting to be undertaken. Penned by a talented wordsmith, this enchanting opus encourages readers on an introspective trip, softly unraveling the veiled truths and profound impact resonating within ab muscles fabric of each word. Within the psychological depths of the poignant review, we shall embark upon a honest exploration of the book is primary themes, dissect its charming publishing design, and fail to the powerful resonance it evokes strong within the recesses of readers hearts.

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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $6\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{CH}_4 + 7\text{O}_2 \rightarrow 2\text{CO}_2 + 8\text{H}_2\text{O}$
 $8\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

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 $1\text{ N}_2 + \text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10}\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow 3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr}_3 + \text{NaBr} + \text{SnO}_4 + \text{NaSO}_3 + \text{Ga}_2\text{SO}_3 + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$ using the equation from problem 2 above answer the following

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 $1\text{ N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $2\text{C}_6\text{H}_{10}\text{O}_2 + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $3\text{HBr} + \text{KHCO}_3 + \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 + \text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr} + 3\text{SnO}_2 + 3\text{NF}_3 + 3\text{SnF}_2 + \text{N}_2\text{O}_3$ using the following equation
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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