

Science Final Study Guide

1. Give at least 3 examples of a chemical reaction.
2. What chemical or physical properties of a substance indicate a chemical reaction may have occurred?
3. How might particles change in a predictable way in regard to motion, temperature, or state of matter after thermal energy is added or removed?
4. How do the total number of atoms change in a chemical reaction in the reactants as opposed to the products? How does the mass and matter change?
5. Explain what would happen when comparing final water temperatures after different masses of ice melted in the same volume of water with the same initial temperature. Include changes in kinetic energy of the particles and temperature.
6. Describe the cycling of water through Earth's systems.
7. What 3 forms does the water take throughout this cycle? Describe them.
8. What propels these changes in forms?

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