

## Polyatomic Ions Worksheet

### Polyatomic Ions

I. Name or write the formula for the following polyatomic ions:

sulfate	_____	$\text{CO}_3^{2-}$	_____
nitrite	_____	$\text{SO}_4^{2-}$	_____
perchlorate	_____	$\text{BrO}_3^-$	_____
hypochlorite	_____	$\text{CO}_3^{2-}$	_____
chlorite	_____	$\text{PO}_4^{3-}$	_____
phosphate	_____	$\text{ClO}^-$	_____
percarbonate	_____	$\text{NO}_2^-$	_____
nitrate	_____	$\text{PO}_4^{3-}$	_____
hyposulfite	_____	$\text{HCO}_3^-$	_____
permanganate	_____	$\text{SO}_4^{2-}$	_____
carbonate	_____		

II. Name or write the formula for the following "Type I" polyatomic ionic compounds:

barium hydrosulfide	_____	$\text{Ba}(\text{OH})_2$	_____
sodium nitrite	_____	$\text{Ca}(\text{OH})_2$	_____
ammonium chloride	_____	$\text{Ag}_2\text{SO}_4$	_____
calcium bisulfate	_____	$\text{MgCO}_3$	_____
rubidium perchlorate	_____	$\text{Sr}_2\text{H}_2\text{O}_4$	_____
chromium sulfate	_____	$\text{Al}(\text{C}_2\text{H}_3\text{O}_5)_3$	_____
aluminum acetate	_____	$\text{SrSO}_4$	_____
ammonium nitrate	_____	$\text{Fe}(\text{OH})_2$	_____
magnesium hypochlorite	_____	$\text{Ca}(\text{HCO}_3)_2$	_____
silver hypochlorite	_____	$\text{Sr}_2\text{CO}_3$	_____
potassium cyanide	_____	$\text{NaNO}_2$	_____
barium iodate	_____	$\text{Ba}(\text{OH})_2$	_____

# Lots Of Ionic Naming Practice Problems Key

**Morris Hein, Scott Pattison, Susan  
Arena, Leo R. Best**



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demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

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**Foundations of College Chemistry** Morris Hein,Susan Arena,Cary Willard,2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically

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Best,2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

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Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level Digital Learning and Teaching in Chemistry Yehudit Dori, Courtney

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