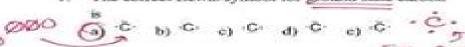
9 • Bonding & Molecular Structure

PRACTICE TEST

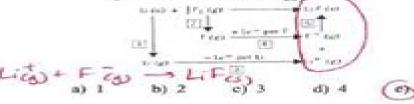
The correct Lewis symbol for ground state carbon



The correct Lewis symbol for ground state advantaments, isc



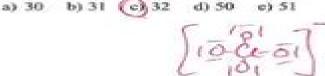
Using the picture below, what process corresponds to the lattice energy?



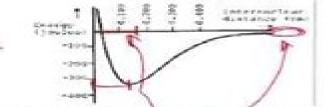
- Which of the following favors formation of an ionic compound?
 - a) low ionization energy for metal
 - b) high electron affinity for non metal
 - c) high lattice energy
 - d)) all of a-c above
 - e) none of a-c above
- Which of the atoms below is least likely to violate. the octet pule?



How many electrons are shown in the Lewis structure of perchlorate ion, ClO₄ ?



Questions7 - 9 refer to the following energy diagram:



What is the energy of the two isolated atoms?

 a) -400 J b) -335 J

e) -155 J

d) 0.100 nm

(c)) 0 J

What is the bond length of the bond between the two atoms?

- a) 0.020 nm
- d) 0.330 mm.

(b)) 0.140 nm

e) 2.00 mm.

c) 0,400 nm

What is the bond energy (bond strength) of the bond? STATE OF THE BE

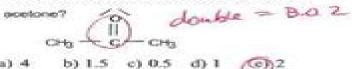
a) 0 J

- b) -110 J

- Transactor

c) -10 J

10. What is the bond order of the C-O bond in



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