



# LM2500 POWER PLANTS

**34 MW**  
SIMPLE CYCLE OUTPUT

**>38%**  
SIMPLE CYCLE EFFICIENCY\*



## CAPABILITY

Ideal for combined cycle, onshore and offshore power generation, mechanical drive, and cogeneration



## VERSATILITY

Accommodates a wide variety of fuels, including coke-oven gas, naphtha, propane, diesel, ethanol, and LNG



## SUSTAINABILITY

Multiple technology options available to lower NO<sub>x</sub> and other emissions concerns

With more than 2,400 units sold and more than 79 million operating hours, GE's LM2500 is the top-selling gas turbine globally. Ideal for CHP applications between 20 to 40 MW, the LM2500's greater than 85 percent efficiency helps reduce operating costs, plant emissions, and reliance on the local grid. A lightweight and compact modular design complements the universal packaging, which improves accessibility for maintenance. The LM2500 delivers power with reliability greater than 99 percent and availability greater than 98 percent.

[gepower.com](http://gepower.com)

		LM2500-DLE	LM2500-LDLE	LM2500-LDLE-SA
SC Plant Performance	SC Net Output (MW)	33.1	33.6	33.6
	SC Heat Rate Ratio (Btu/kWh, LHV)	11,501	11,442	11,497
	SC Heat Rate Ratio (Btu/kWh, HHV)	12,854	12,482	12,381
	SC Plant Efficiency (% LHV)	39.9%	40.0%	40.1%
LC Plant Performance	CC Net Output (MW)	52.2	45.9	47.2
	CC Heat Rate Ratio (Btu/kWh, LHV)	8,465	8,789	8,719
	CC Heat Rate Ratio (Btu/kWh, HHV)	9,811	9,545	9,565
	CC Plant Efficiency (% LHV)	54.0%	54.2%	54.3%
	Plant Turn-down – Minimum Load (%)	34.0%	35.0%	35.0%
	Plant Water (MG/Week)	30	30	30
	Startup Time (20% Hot, Minimum)	30	30	30
2x1 Plant Performance	CC Heat Output (MW)	88.9	89.4	89.1
	CC Heat Rate Ratio (Btu/kWh, LHV)	6,411	6,271	6,235
	CC Heat Rate Ratio (Btu/kWh, HHV)	6,785	6,622	6,560
	CC Plant Efficiency (% LHV)	61.1%	61.4%	61.5%
	Plant Turn-down – Minimum Load (%)	17.0%	17.0%	18.0%
	Plant Water (MG/Week)	60	60	60
Startup Time (20% Hot, Minimum)		30	30	30

\*Based on ratings of combined cycle at 50% efficiency and power output. Actual performance will vary with plant specific conditions and fuel.

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GE A 300419A 1A 101 (09/16)

# Ge Lm2500 Plus Technical Manual

**Puja Mehta**



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web oct 6 2022 therefore boyle s law indicates that when the pressure of the object increases the volume of the object decreases and vice versa because volume and pressure s relationship is inversely proportional while the temperature is constant

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web overall the experiment succeeded in showing that temperature and pressure for an ideal gas at constant volume and mass follow the relation of the ideal gas equation differences existed in the experimental graph of temperature versus and pressure and the theoretical curve of temperature versus pressure



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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web chm 267 lab 7 lab report preview text pressure temperature relationship in gases lab shannon urmetz 2702902 chem 266 section 01 introduction in this lab we observed the relationship between pressure and temperature

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 20 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web chemistry lab 117 experiment 3 pressure volume and temperature discovery of the gas laws lab partner amazona tahbou october 8 2014 summary the purpose of this experiment is to be able to understand the relationship between the pressure volume and temperature of a gas understanding this relationship is important as it allows

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the

standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law  $pV = nRT$  is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 K and 1 atm or 101 325 kPa to report properties of gases

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web the pressure volume temperature pvt fluids lab in ku's chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web volume pressure kPa linear pressure kPa volume mL pressure kPa in part b we observed the relationship between temperature and pressure the pressure of the 125 mL flask was at 101 36 kPa when the temperature was at 23 2 °C

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kPa temperature °C temperature K constant k p t 95 82 kPa 16 0 °C 289 K 0 33 97 90 kPa 19 0 °C 292 K 0 34 98 97 kPa 99 22 kPa 101 4 kPa 19 1 °C

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts

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