

# TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm<sup>3</sup>

Use only the student's concordant results

mean titre =  $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$  ✓

$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol}$  ✓

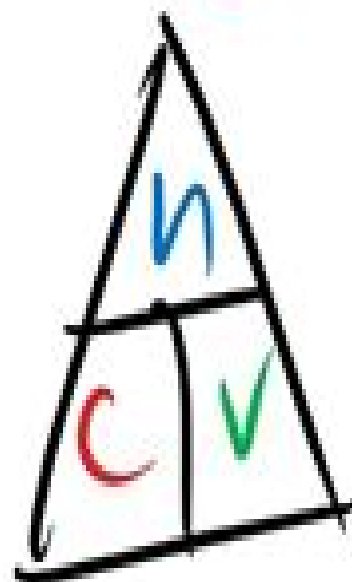
$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol}$  ✓

$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3$  ✓

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



# GCSE CHEMISTRY

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**Michael Seilmaier**

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Gcse Titration Practise Questions**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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