



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Joacim Rocklöv



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Table of Contents Memo Physical Science Grade11 Controlled Test 2 Term3

1. Understanding the eBook Memo Physical Science Grade11 Controlled Test 2 Term3
 - The Rise of Digital Reading Memo Physical Science Grade11 Controlled Test 2 Term3
 - Advantages of eBooks Over Traditional Books
2. Identifying Memo Physical Science Grade11 Controlled Test 2 Term3
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Memo Physical Science Grade11 Controlled Test 2 Term3
 - User-Friendly Interface
4. Exploring eBook Recommendations from Memo Physical Science Grade11 Controlled Test 2 Term3
 - Personalized Recommendations
 - Memo Physical Science Grade11 Controlled Test 2 Term3 User Reviews and Ratings
 - Memo Physical Science Grade11 Controlled Test 2 Term3 and Bestseller Lists
5. Accessing Memo Physical Science Grade11 Controlled Test 2 Term3 Free and Paid eBooks
 - Memo Physical Science Grade11 Controlled Test 2 Term3 Public Domain eBooks
 - Memo Physical Science Grade11 Controlled Test 2 Term3 eBook Subscription Services
 - Memo Physical Science Grade11 Controlled Test 2 Term3 Budget-Friendly Options
6. Navigating Memo Physical Science Grade11 Controlled Test 2 Term3 eBook Formats

- ePub, PDF, MOBI, and More
 - Memo Physical Science Grade11 Controlled Test 2 Term3 Compatibility with Devices
 - Memo Physical Science Grade11 Controlled Test 2 Term3 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Memo Physical Science Grade11 Controlled Test 2 Term3
 - Highlighting and Note-Taking Memo Physical Science Grade11 Controlled Test 2 Term3
 - Interactive Elements Memo Physical Science Grade11 Controlled Test 2 Term3
 8. Staying Engaged with Memo Physical Science Grade11 Controlled Test 2 Term3
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Memo Physical Science Grade11 Controlled Test 2 Term3
 9. Balancing eBooks and Physical Books Memo Physical Science Grade11 Controlled Test 2 Term3
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Memo Physical Science Grade11 Controlled Test 2 Term3
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Memo Physical Science Grade11 Controlled Test 2 Term3
 - Setting Reading Goals Memo Physical Science Grade11 Controlled Test 2 Term3
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Memo Physical Science Grade11 Controlled Test 2 Term3
 - Fact-Checking eBook Content of Memo Physical Science Grade11 Controlled Test 2 Term3
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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