

4. A 50.0-g sample of Mg(OH)_2 is reacted with 45.0 g of HCl producing magnesium chloride and water. Identifying the limiting reagent. What is the theoretical yield of magnesium chloride?
5. Methane, CH_4 , burns in oxygen to give carbon dioxide and water. In one experiment, a mixture of 2.250 mol of methane was burned in 1.25 mol of oxygen in a sealed steel vessel. Find the limiting reagent, if any, and calculate the theoretical yield, (in moles) of water.
6. Chloroform, CHCl_3 , reacts with chlorine to form carbon tetrachloride and hydrogen chloride. In an experiment 55 g of chloroform and 12 g of chlorine were mixed. Which is the limiting reagent? What is the theoretical yield of CCl_4 in grams?

Limiting Reactant Problems And Answers High School

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