Second Semester Final Exam

Multiple Choice: Please mark the answer to the multiple choice questions on your bubble sheet.

- The molar mass of magnesium cyanide, Mg(CN)₂, is approximately
 - A) 50.3 g
 - B) 64.3 g
 - C) 76.3 g
 - D) 100.6 g
- The total number of atoms in (NH₄)₃BO₃ is
 - A) 12
 - B) 17
 - C) 19
- 3) What is the mass in grams of 5.90 mol C₈H₁₈?
 - A) 0.0512 g
 - B) 389 g
 - C) 673 g
 - D) 19.4 g
- 4) How many atoms are there in 1 mole of helium?
 - A) 6.02 x 10²³ atoms
 - B) 22.4 atoms
 - C) 4.0 atoms
 - D) 0.0821 atoms
- The percent composition of silver (Ag) in AgNO₃ is
 - A) 0:625 %
 - B) 1.57.%
 - C) 62.5%
 - D) 157 %
 - E) 169.9 %
- 6) Which of the following is an empirical formula?
 - A) C₂H₄
 - B) C₂H₆
 - C) CH₄
 - D) CaHa
- Which of the following could be a molecular formula for the empirical formula CH₂
 - A) CaHa
 - B) CaHa
 - C) CH
 - D) CH4
- 8) What is the coefficient in front of the O₂ when the following reaction is correctly balanced?

 $C_1H_8 + O_2 \rightarrow CO_2 + H_2O$

- A) 1 B) 2
- C) 3
- (D) 4
- HD 5

<u>Ied 2nd Semester Final Exam Answers</u>

Clemens Wendtner

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