

- 10 In 1859, Thomas Edison invented the nickel-iron battery. The following reaction takes place in the battery.



What is the limiting reactant when 5.00 mol of Fe and 8.00 mol of $\text{Ni}(\text{OH})_2$ react?

How many moles of $\text{Fe}(\text{OH})_2$ will be produced?

How many moles of the other reactant will be in excess?

- 20 One of the first cerium compounds that forms is cerium sesquifluoride (Ce_2CeF_7).



What is the limiting reactant in the reaction of 12.5 mol of cerium fluoride with 50.0 mol of sesquifluoride?

How many moles of Ce_2CeF_7 can be produced?

How many moles of the other reactant will be excess?

- 30 Iron is obtained commercially by the reaction of hematite (Fe_2O_3) with carbon monoxide.



What is the limiting reactant when 25.0 mol of hematite reacts with 30.0 mol of carbon monoxide?

How many grams of Fe can be produced?

Limiting Reactant And Percent Yield Packet

Liying Dong



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Chemistry Rob Lewis, Wynne Evans, 2018-03-31 The fifth edition of this engaging and established textbook provides students with a complete course in chemical literacy and assumes minimal prior experience of science and maths Written in an accessible and succinct style this book offers comprehensive coverage of all the core topics in organic inorganic and physical chemistry Topics covered include bonding moles solutions and solubility energy changes equilibrium organic compounds and spectroscopy Each unit contains in text exercises and revision questions to consolidate learning at every step and is richly illustrated with diagrams and images to aid understanding This popular text is an essential resource for students who are looking for an accessible introductory textbook It is also ideal for non specialists on courses such as general science engineering environmental health or life sciences New to this Edition A foreword by Professor Sir John Meurig Thomas FRS former Director of the Royal Institution Three additional units on Gibbs Energy Changes Organic Mechanisms and Fire and Flame

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