



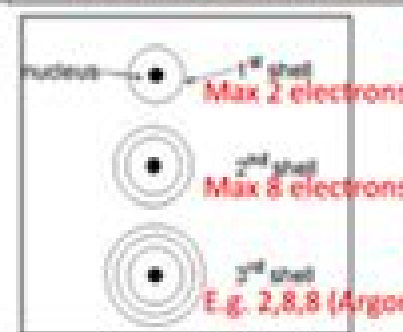
Periodic Table of the Elements

An atom:
 Number of Protons = Number of Electrons
 Mass number = Protons + Neutrons

C2 Chemistry

Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
 E.g. Lithium 2,1 is a group 1 element.

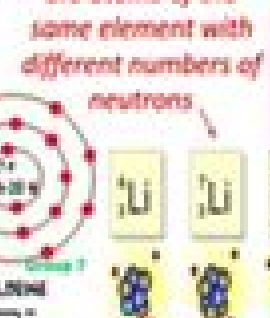
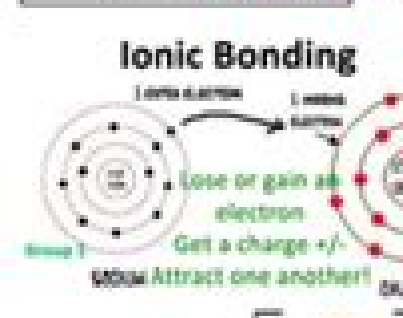
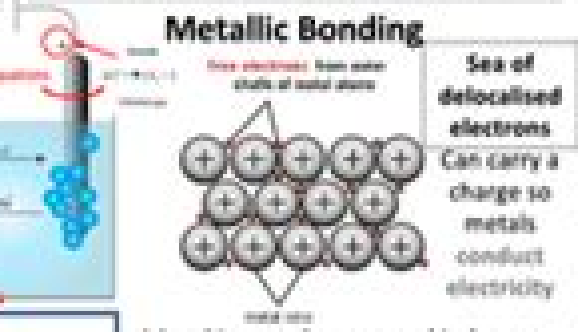
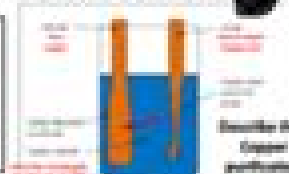


Protons and neutrons both have a mass of 1

ISOTOPES are atoms of the same element with different numbers of neutrons

OIL RIG

Oxidation is loss (of electrons)
 Reduction is gain (of electrons)

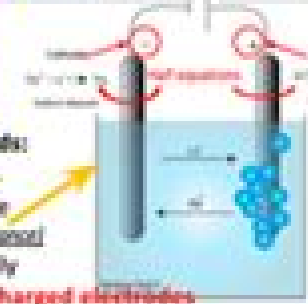


Electrolysis:
 Splitting up a substance using electricity

Electrolysis needs:

- 2 electrodes
- An electrolyte (ionic or molten substance)
- A power supply

When there are 2 possible products The less reactive forms at the electrode

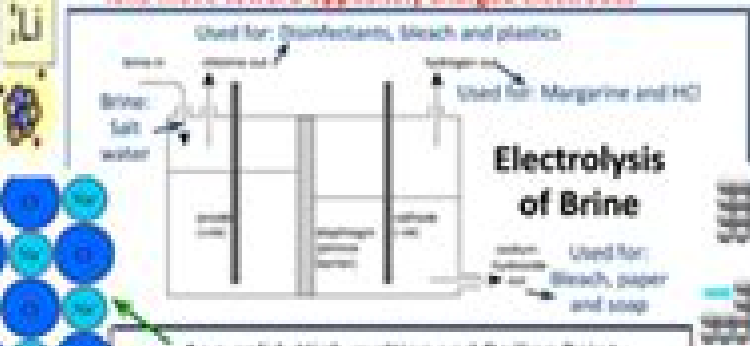


Metal ions only arranged in layers

ELECTROSTATIC ATTRACTION between +IONS and ELECTRONS

Metal ions slide over one another making them **MALLEABLE**

Metals conduct **HEAT** well



NANO-TECHNOLOGY
 Means **REALLY REALLY REALLY** small!!!!

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1. Understanding the eBook Gcse Edexcel Chemistry C2 Past Papers
 - The Rise of Digital Reading Gcse Edexcel Chemistry C2 Past Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gcse Edexcel Chemistry C2 Past Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gcse Edexcel Chemistry C2 Past Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gcse Edexcel Chemistry C2 Past Papers
 - Personalized Recommendations
 - Gcse Edexcel Chemistry C2 Past Papers User Reviews and Ratings
 - Gcse Edexcel Chemistry C2 Past Papers and Bestseller Lists

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 - Highlighting and Note-Taking Gcse Edexcel Chemistry C2 Past Papers
 - Interactive Elements Gcse Edexcel Chemistry C2 Past Papers
8. Staying Engaged with Gcse Edexcel Chemistry C2 Past Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gcse Edexcel Chemistry C2 Past Papers
9. Balancing eBooks and Physical Books Gcse Edexcel Chemistry C2 Past Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gcse Edexcel Chemistry C2 Past Papers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gcse Edexcel Chemistry C2 Past Papers
 - Setting Reading Goals Gcse Edexcel Chemistry C2 Past Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gcse Edexcel Chemistry C2 Past Papers
 - Fact-Checking eBook Content of Gcse Edexcel Chemistry C2 Past Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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