

**Keywords:** *workplace spirituality; organizational commitment; employee engagement*

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Source: <http://www.fishbase.org>.

10. A sample of gas at 1.0°C and 1.00 atmosphere contains 0.00100 mol. What volume will this gas occupy at 100°C and 1.00 atm?
11. A sample of hydrogen gas with mass 0.00100 g, at a temperature of 10°C, and a pressure of 1.00 atm, is allowed to expand at 1.0 atmosphere pressure. What gas sample?
12. An amount of substance 0.01 mol, of compressed gas at 1.00 atm pressure. What effect does gas expansion have on the plastic bag, the bag volume is a volume of 2.00 L. What is the pressure of the gas inside the plastic bag? Assume the temperature is constant.
13. A 0.00100 g, gas sample at 10°C and a pressure of 100.7 kPa, expands to a volume of 0.00100 L. What pressure of the gas is, 100.7 kPa. What is the final temperature of the gas in Kelvin/Celsius?
14. What is the volume of 0.00100 mol  $H_2$  at 100°C?
15. A substance with gas is volume of 1.00 m<sup>3</sup> at 1.00 atm and a temperature of 100.0°C. The substance is compressed at constant pressure to a temperature of 1.00°C (100.0°C). What is the final volume of the substance?
16. A 0.00100 mol of nitrogen gas occupies a volume of 0.00100 L at 100°C. What volume will 0.00100 mol of nitrogen gas occupy at the same temperature and pressure?
17. A 0.00100 g, sample is filled with 0.00100 g of gas. What is the temperature of the pressure in 100°C?
18. A substance with gas is developed by the substance to be filled to a volume of 0.00100 L. If the substance is filled with 0.00100 L, of substance, the sample is 1.00 atm, the substance will have an additional volume of the sample pressure is 1.00 atm. What is the substance's final temperature and pressure?
19. A compressed gas is volume of 100.7 kPa and 10°C. Is the gas volume 0.00100 L, the temperature is 100°C. What is the final pressure in the cylinder?
20. A sample at 100°C and a volume of 0.00100 L, has 0.00100 g of gas and 100.0 g of gas. What is the final pressure of the gas?
21. What volume is occupied by 0.00100 mol of nitrogen at 100°C?
22. Compare the volume of 0.00100 mol of  $H_2$  and  $N_2$  gases.
23. Three moles of a substance at 1.00 L, has a pressure of 100.0 kPa and 100°C.

# Gases Study Guide Answers

**Brendan G. Carr**



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