





**Student Directions for Gas Properties** Chemistry: Gas Laws  
<http://phet.colorado.edu/>

**Learning Goals:**

- Design experiments to measure the relationships between pressure, volume, and temperature.
- Create graphs based on predictions and observations.
- Make qualitative statements about the relationships between pressure, volume and temperature using molecular models.

**Predictions:** Make a chart like the one below. Without using the simulation, sketch what you think the graphs would look like. **Note: Be sure to label your x and y axes.**

<p>I. Volume-Pressure graph</p> 	<p>Explain your reasoning for the graph's appearance</p>
<p>II. Volume-Temperature graph</p> 	<p>Explain your reasoning for the graph's appearance</p>
<p>III. Temperature-Pressure graph</p> 	<p>Explain your reasoning for the graph's appearance</p>
<p>IV. Number of particles – Volume</p> 	<p>Explain your reasoning for the graph's appearance</p>

**Experiments:**

1. For each case, I-IV. Write a short description of how to use the sim to collect data. Then make an Excel spreadsheet for each, graph and curve fit. Some helpful hints – if you set a parameter like temperature constant, then make a change, you

# Gas Properties Phet Lab Answers

**American Gas Association (AGA)**



## Gas Properties Phet Lab Answers:

**Even More Brain-powered Science** Thomas O'Brien, 2011 The third of Thomas O'Brien's books designed for 50 Co12 grade science teachers Even More Brain Powered Science uses questions and inquiry oriented discrepant events or experiments or demonstrations in which the outcomes are not what students expect or to dispute misconceptions and challenge students to think about discuss and examine the real outcomes of the experiments O'Brien has developed interactive activities many of which use inexpensive materials or to engage the natural curiosity of both teachers and students and create new levels of scientific understanding **Gas Properties**, *Gas Properties*, **Gas Properties**,

**Ideal Gas Properties** Jing Chao, 1986 Gas Properties and Compressor Data Ingersoll-Rand Company, 1966

Present State of Our Knowledge of Gas Properties, 1947 Gas Measurement Manual: Measurement of gas properties American Gas Association, 1977 COMPENDIUM OF GAS PROPERTIES A-6C3 Fluids Committee, 2008 *Experimental Determination of Gas Properties at High Temperature And/or Pressures* C. Carey, 1974 **Gas Measurement Manual** American Gas Association (AGA), 1988 Gas Properties Lefebvre, 2019 This lesson plan covers gas compression gas pressure and using a vapor pressure curve to determine boiling points at different atmospheric pressures *Experimental Determination of Gas Properties at High Temperatures And/or Pressures* C. Carey, E. H. Carnevale, Arnold Engineering Development Center, S. Uva, T. Marshall, PANAMETRICS INC WALTHAM MASS., 1968 The problem of extending transport property measurements to conditions of high pressure **Teaching the Gas Properties and Gas Laws** Michael T. Hammar, 2013 As a chemistry teacher I felt that the gas laws were rather simple concepts to understand My students however felt the opposite Conceptual issues as well as mathematical issues were problematic every year In this study a unit about gas properties and gas laws was modified to include inquiry based teaching methods The research questions focused on how these changes affected student results on a traditional end of the unit test and on an alternative assessment Students attitudes during these lessons were also monitored to see if the inquiry methods improved student perceptions of teaching learning and self awareness The results of this study showed that an inquiry approach improved the student's ability to perform on a traditional end of the unit test in the areas of microscopic understanding atomic level symbolic understanding mathematical level and graphical understanding relationships between pressure volume and temperature It was also shown to have a positive effect on the students confidence and attitude towards the material presented when inquiry was utilized On the other hand the revised unit did not improve the students overall performance on an alternative assessment geared towards the concept of pressure but showed some improvement in two specific areas a students understanding that pressure is created by molecules and their movement and b students understanding that when pressure is equalized little happens within a system **The Physical Properties of Gases** Arthur Lalanne Kimball, 1890 **Combustion Gas Properties**, 1985 *Tables of Thermal Properties of Gases* Joseph Hilsenrath, Charles W. Beckett, William S. Benedict, 2013-03

Additional Contributors Are Lilla Fano Harold J Hoge Joseph F Masi Ralph L Nuttall Yeram S Touloukian And Harold W Woolley Preface By A V Astin     The Properties of Gases Marylou Morano-Kjelle, 2006-07-15 Introduces the physical properties of gases and discusses why oxygen nitrogen and hydrogen are such important gases     *Combustion Gas Properties* ,1985     Combustion Gas Properties ,1985

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