# **Experiment 5-Worksheet Nomenclature**

Write formula or name for each compound:

# IA. Binary ionic compounds

	Name	Formula	Formula	Name
1	sodium oxide	Direct 1	BaS	
2	calcium nitride		AlBra	
3	barium iodide		Ca <sub>3</sub> P <sub>2</sub>	
4	potassium sulfide		Mg <sub>3</sub> N <sub>2</sub>	
5	aluminum oxide		K₂O	

# IB. Transition metals with one oxidation state

	Name	Formula	Formula	Name
1	silver oxide		ZnCl <sub>2</sub>	
2	zinc phosphide		AgI	
3	zinc sulfide		AgsP	

#### IC. Metals more than one oxidation state

	Name	Formula	Formula	Name
1	cupric oxide		FeBra	
2	iron (III) sulfide		Cul	
3	ferrous nitride		CuS	

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**Boynton Wells McFarland** 

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#### Table of Contents Formulas And Nomenclature Oxidation Lab

- 1. Understanding the eBook Formulas And Nomenclature Oxidation Lab
  - The Rise of Digital Reading Formulas And Nomenclature Oxidation Lab
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Formulas And Nomenclature Oxidation Lab
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Formulas And Nomenclature Oxidation Lab
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Formulas And Nomenclature Oxidation Lab
  - Personalized Recommendations
  - Formulas And Nomenclature Oxidation Lab User Reviews and Ratings
  - Formulas And Nomenclature Oxidation Lab and Bestseller Lists
- 5. Accessing Formulas And Nomenclature Oxidation Lab Free and Paid eBooks

- Formulas And Nomenclature Oxidation Lab Public Domain eBooks
- Formulas And Nomenclature Oxidation Lab eBook Subscription Services
- Formulas And Nomenclature Oxidation Lab Budget-Friendly Options
- 6. Navigating Formulas And Nomenclature Oxidation Lab eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Formulas And Nomenclature Oxidation Lab Compatibility with Devices
  - Formulas And Nomenclature Oxidation Lab Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Formulas And Nomenclature Oxidation Lab
  - Highlighting and Note-Taking Formulas And Nomenclature Oxidation Lab
  - Interactive Elements Formulas And Nomenclature Oxidation Lab
- 8. Staying Engaged with Formulas And Nomenclature Oxidation Lab
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Formulas And Nomenclature Oxidation Lab
- 9. Balancing eBooks and Physical Books Formulas And Nomenclature Oxidation Lab
  - Benefits of a Digital Library
  - $\circ$  Creating a Diverse Reading Collection Formulas And Nomenclature Oxidation Lab
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Formulas And Nomenclature Oxidation Lab
  - Setting Reading Goals Formulas And Nomenclature Oxidation Lab
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Formulas And Nomenclature Oxidation Lab
  - Fact-Checking eBook Content of Formulas And Nomenclature Oxidation Lab
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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